

Binap-AgSbF₆ versus Binap-AgClO₄ complexes as catalysts for the enantioselective 1,3-dipolar cycloaddition of azomethine ylides and alkenes

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Dedicated to Prof. Antonio García Martínez on the occasion of his retirement

Abstract: The employment of AgSbF₆ and Binap ligands has been evaluated in the catalyzed enantioselective 1,3-dipolar cycloadditions (1,3-DC) between azomethine ylides and electrophilic alkenes. The results are compared with the analogous ones obtained when using AgClO₄. The cycloaddition with maleimides and *trans*-1,2-bis(phenylsulfonyl)ethylene are clearly improved by the AgSbF₆ derived catalyst, and its efficiency is crucial for ensure good yields and excellent *ee* in the three-component reaction.

Key words: asymmetric catalysis / prolines / silver salts / azomethine ylides / multicomponent reaction

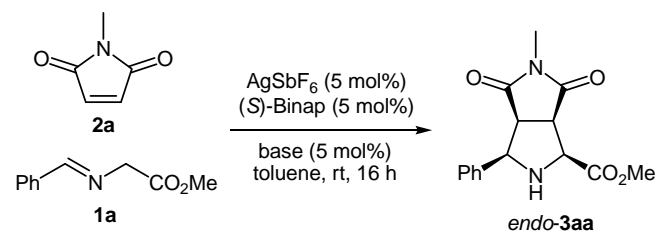
The 1,3-dipolar cycloaddition¹ (1,3-DC) became a very important transformation since the appearance in 2002² of the first catalytic enantioselective transformation employing substoichiometric amounts of a silver(I) chiral complex.^{3,4} The importance of this reaction is due to the generation of up to four stereogenic centers in the resulting attractive proline derivatives.⁵ It is noteworthy that it is produced in only one reaction step using very soft reaction conditions. To date, this regio-, diastereo- and enantioselective transformation becomes more efficient when chiral Lewis acids [obtained from silver(I),^{6,7} copper(I),⁸ zinc(II),⁹ nickel(II)¹⁰, and calcium(II)¹¹ salts] are employed. For example, the wide scope of both of the iminoesters and dipolarophiles, without structural limitations and the lower catalyst loading employed, are the main advantages of this catalysis *versus* those organocatalysed¹² processes.¹³

In a recent publication, we described the Binap-AgClO₄ catalyzed 1,3-DC between azomethine ylides and maleimides.⁷ The high stability of the titled catalytic metal-complex to light exposure and its insolubility in toluene made possible its recovery and reutilization. Additionally, it was found that the transition state responsible of the enantiodiscrimination was quite asynchronous, the perchlorate anion being weakly bonded to the central metal.^{7b}

According to this precedent, we envisaged that the poorly coordinating anion SbF₆⁻ would modify the chiral domain of the metal complex vacancy. So, in this communication we will describe the improvements achieved by the use of AgSbF₆ in the 1,3-DC of azomethine ylides and alkenes employing (*R*)- or (*S*)-Binap as chiral ligand *versus* the already detailed results obtained with the perchlorate anion.

The standard reaction between methyl benzylidene-iminoglycinate **1a** and *N*-methylmaleimide (NMM) **2a** (Table 1) was employed for the optimization tests. Initially, the reaction conditions were identical to those previously described for Binap-AgClO₄ (Table 1, entry 1) catalyzed 1,3-DC, that means, 5 mol% of the catalyst (formed with equimolar amounts of chiral Binap and AgSbF₆), in toluene at room temperature for 16 h, in the presence of Et₃N as base. Both of the AgClO₄ and AgSbF₆ catalyzed reactions gave identical results of the cycloadduct *endo*-**3a** (90% yield, >98:2 *endo:exo* ratio, and >99% *ee*) (Table 1, entries 1 and 2). Other bases like 1,4-diazabicyclo[2.2.2]octane (DABCO) or diazabicyclo[5.4.0]undec-7-ene (DBU) were evaluated in this reaction affording lower enantioselections with some unidentified secondary products in the crude product (Table 1, entries 3, and 4). As well as described for the catalyst formed from AgClO₄,⁷ a different stoichiometry such as 2:1 or 1:2 chiral ligand:silver salt, afforded lower enantioselections and larger amounts of the *exo*-diastereoisomer (Table 1, entries 5 and 6). Additionally, other solvents such as THF, DCM, or Et₂O did not improve neither yields nor enantiomeric excesses. In general, a similar behavior was observed for both of the silver complexes for all the experiments summarized on Table 1.

Table 1. Reaction conditions study of the reaction of **1a** and **2a** using 1:1 mixture of (*S*)-Binap and AgSbF₆ or AgClO₄



	(S)-Binap:		product 3aa ^a	
	Ag salt	base	yield (%) ^b	<i>ee</i> _{endo} (%) ^c
1	1:1	Et ₃ N	90 (89)	>99 (>99)
2	1:1 ^d	Et ₃ N	78 (80)	>99 (>99)
3	1:1	DABCO	87 (88)	94 ^e (92) ^e
4	1:1	DBU	87 (80)	90 ^e (91) ^e

5	2:1	Et ₃ N	90 (90)	96 ^f (98) ^f
6	1:2	Et ₃ N	90 (91)	56 ^{e,f} (<50) ^{e,f}

^a The *endo:exo* ratio was always >98:2 (¹H NMR spectroscopy, and chiral HPLC).

^b Isolated after recrystallization using AgSbF₆, in brackets the yields of the analogous reaction performed with AgClO₄.⁷

^c Determined by chiral HPLC (Daicel Chiralpak AS) of the crude product, in brackets the *ee* of the analogous reaction performed with AgClO₄.⁷ Identical *ee* was determined after purification of **3a**.

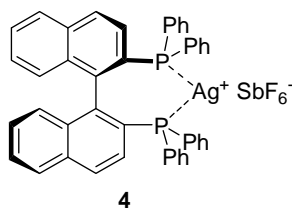
^d Reaction performed using 3 mol% of catalyst.

^e Some unidentified impurities were observed in the crude product.

^f The *endo:exo* ratio was approximately 90:10.

The reaction was also carried out with the isolated complex formed by the addition of equimolar amounts (*S*)-Binap and AgSbF₆, obtaining almost identical results to those described in entry 1 of the Table 1. However this AgSbF₆ derived complex became darker upon standing being much more unstable than the identical complex generated by AgClO₄. So, the *in situ* generation of the catalytic complex, avoiding the light exposure during the whole process was preferred for all the transformations described in the text.

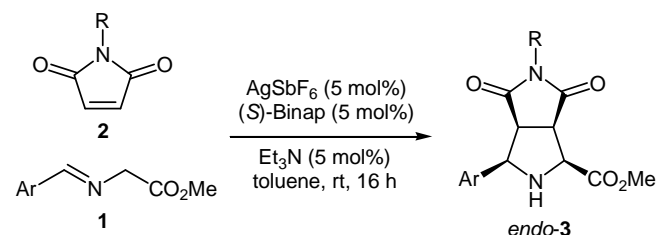
The presumed catalytic monomeric species in solution are identical to those reported previously with different anions.^{7,14} In fact, the 1:1 (*R*)- or (*S*)-Binap and AgSbF₆ complexes were characterized by ESI-MS experiments and ³¹P NMR. ESI-MS revealed a M⁺+1 signal at 730, and 732 corresponding to the monomeric Binap-Ag^I complex (*S*)-**4** and a tiny one at 1352 and 1354 corresponding to the 2:1 Binap:AgSbF₆.^{7,15} The ³¹P NMR (CDCl₃) spectra of 1:1 (*R*)- or (*S*)-Binap and AgSbF₆ (10% aqueous polyphosphoric acid as internal reference) afforded signals at 15.31 ppm and 15.45 ppm (2d, *J* = 242 Hz) (15.26, and 15.35 ppm for Binap-AgClO₄ complex). The absence of NLE is another data supporting the existence of a monomeric species in the enantioselective catalytic process in solution.



The reaction of several azomethine ylides, generated from the corresponding methyl arylideneiminoglycinates **1**, and maleimides in toluene employing a 5 mol% of an equimolar mixture of (*S*)-Binap and AgSbF₆ was studied and compared with the analogous results obtained with (*S*)-Binap-AgClO₄ complex⁷ (Table 2). In general, we can observe that isolated chemical yields of compounds **3** are identical to each other finding a higher enantioselectivity in those reactions promoted by the complex formed by (*S*)-Binap and AgSbF₆. This is more noticeable when the aromatic moiety was substituted at different positions as, for example, cycloadducts **3ba**, **3ca**, **3da**, and **3ea** (Table 2, entries 3-6). Whilst the reaction with *N*-ethylmaleimide (NEM) do not represent any differ-

ence with respect to those results obtained in the (*S*)-Binap-AgClO₄ (Table 2, entry 7), the reaction carried out with *N*-phenylmaleimide (NPM) was much more enantioselective in the presence of (*S*)-Binap-AgSbF₆ complex (82% *ee*, vs 62% *ee* obtained with perchlorate derived chiral complex) and >98:2 *endo:exo* ratio was obtained (Table 2, entry 8). Computational calculations have revealed the existence of steric hindrance between the phenyl group of the NPM and the Binap-AgClO₄ catalyst.^{7b} However, in the case of Binap-AgSbF₆ catalyzed process seems that the less coordinating anion decreases the steric congestion of the transition state.

Table 2. 1,3-DC of maleimides and several iminoglycinates **1**.



	Ar	R	3 ^a	yield (%) ^{b,c}	<i>ee</i> _{endo} (%) ^{c,d}
1	Ph	Me	3aa	90 (90)	>99 (>99)
2	Ph ^e	Me	<i>ent</i> - 3aa	90 (90)	>99 (>99)
3	2-MeC ₆ H ₄	Me	3ba	85 (85)	99 (70)
4	2-ClC ₆ H ₄	Me	3ca	82 (82)	>99 ^f (85) ^f
5	4-MeC ₆ H ₄	Me	3da	85 (88)	99 (88)
6	4-MeOC ₆ H ₄	Me	3ea	85 (85)	99 (80)
7	Ph	Et	3ab	84 (91)	99 (99)
8	Ph	Ph	3ac	86 (86)	82 (62) ^g

^a The *endo:exo* ratio was always >98:2 (¹H NMR spectroscopy, and chiral HPLC).

^b Isolated after flash chromatography.

^c In brackets the result obtained previously with (*S*)-Binap-AgClO₄ complex.⁷

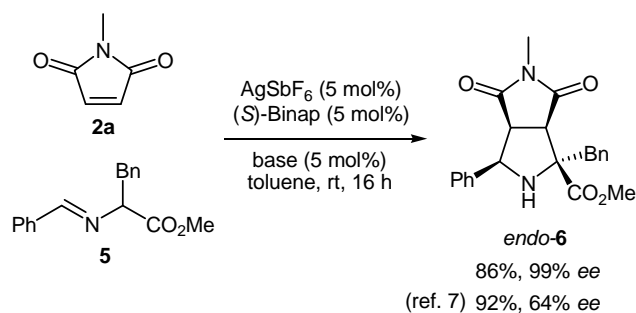
^d Determined by chiral HPLC of the crude product. Identical *ee* was determined after purification of **3**.

^e Reaction performed with (*R*)-Binap-AgSbF₆.

^f Reaction performed at -20 °C.

^g For the AgClO₄-catalyzed process, the *endo:exo* ratio was approximately 90:10.

The incorporation of a bulky substituent at the α-position of the 1,3-dipole precursor as occurred in the methyl benzylideneiminophenylalaninate **5** was successfully overcome using the standard reaction conditions. Here, the *endo*-adduct **6** was obtained as a single enantiomer (99% *ee*) in very good chemical isolated yield (86%). Clearly this reaction promoted by (*S*)-Binap-AgSbF₆ complex improve the results obtained in the reaction performed with the perchlorate salt (Scheme 1), when steric problems are presented in the 1,3-DC.



Scheme 1

Acrylates, maleates and fumarates were not suitable dipolarophiles neither for the (*S*)-Binap-AgSbF₆ (<30% *ee* and <40% *ee*, respectively) nor for (*S*)-Binap-AgClO₄⁷ catalysed processes. However *trans*-1,2-bis(phenylsulfonyl)ethylene **7** afforded very interesting results of *endo*-cycloadducts **8**. This electrophilic alkene is a synthetic equivalent of acetylene and has been employed in the synthesis of the corresponding *exo*-cycloadducts in the presence of chiral Fesulphocopper(I) complex following similar reaction conditions.^{8d} The reaction operates under the standard conditions but taking 48 h to complete. This 1,3-DC, not evaluated previously with the chiral perchlorate complex, was performed using both silver salts (Table 3). The four methyl iminoglycinates **1** tested in this reaction under the control of (*S*)-Binap-AgSbF₆ catalytic complex afforded *endo*-cycloadducts **8** in good chemical yields (80-91%) and very high enantioselectivities (88-92% *ee*) (Table 3). The enantioselection was higher for AgSbF₆ by the chiral silver complex in all of the entries described in Table 3, especially in the reaction of *p*-tolyliminoester (Table 3, entry 3). This is the first synthesis of the disulfonyl *endo*-**8** cycloadducts, whose absolute configuration was determined by NOESY experiments, and indirectly by comparison of the corresponding HPLC analysis with those described in the literature for the *exo*-cycloadducts.^{8d}

Table 3. 1,3-DC of (*E*)-1,2-bis(phenylsulfonyl)ethylene **7** and iminoglycinates **1**.



	Ar	8 ^a	yield (%) ^{b,c}	<i>ee</i> _{endo} (%) ^{c,d}
1	Ph	8a	81 (80)	90 (88)
2	Ph ^e	<i>ent</i> - 8a	80 (80)	90 (88)
3	4-MeC ₆ H ₄	8b	91 ^f (85)	88 (28)
4	3-Pyridyl	8c	83 (82)	93 (78)

5 2-Naphthyl **8d** 91^f (88) 92 (80)

^a The *endo:exo* ratio was >98:2 (¹H NMR spectroscopy, and chiral HPLC).

^b Isolated after column chromatography.

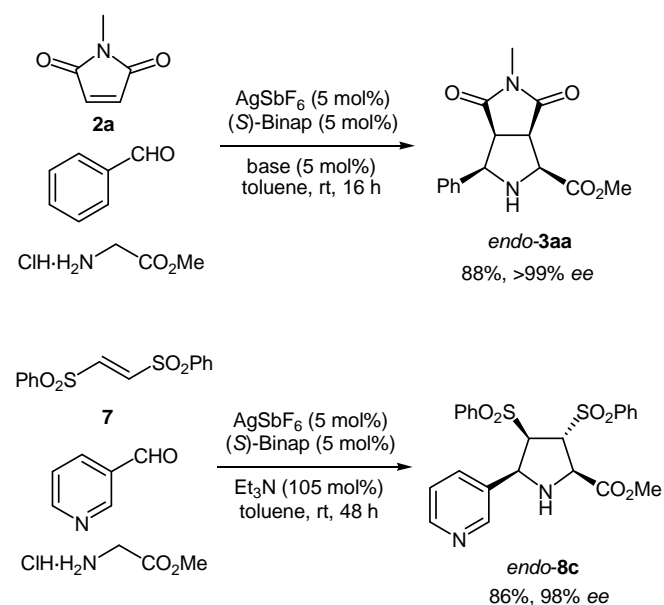
^c In brackets the result obtained with (*S*)-Binap-AgClO₄ complex.

^d Determined by chiral HPLC of the crude product. Identical *ee* was determined after purification of **8**.

^e Reaction performed with (*R*)-Binap-AgSbF₆.

^f Pure crude yields.

The multicomponent reactions were attempted using the best result depicted in Tables 2 and 3. Thus, benzaldehyde/NMM or 3-pyridinecarbaldehyde/disulfone **7**, glycine methyl ester hydrochloride, triethylamine (1.05 equiv), (*S*)-Binap-AgSbF₆ (5 mol%), were put together in toluene and the resulting mixture was allowed to react at rt for 48 h. The results obtained for compound *endo*-**3aa** or *endo*-**8c** were impressive (88% yield, >99% *ee*, or 86% yield, 98% *ee*, respectively, Scheme 2). However, taking in account that the analogous reactions in the presence of (*S*)-Binap-AgClO₄ complex failed. Although very activated aminomalonates have been involved as one of the three components of the enantioselective organocatalyzed 1,3-DC,^{12c,i,j,k} this is the first occasion that a three-component transformation is enantioselectively performed in the presence of a chiral Lewis acid.



Scheme 2

The high enantioselectivity observed for the reaction using different maleimides, and 1,2-bis(phenylsulfonyl)ethylene as dipolarophiles with (*S*)-Binap-AgSbF₆ catalytic complex¹⁶ suggest a weaker coordination of the SbF₆ anion to the metal centre. The enantioselections, in general, are higher with this novel chiral catalyst constituting this study a clear example of fine tuning of the catalyst for the obtention of highly enantiomerically enriched *endo*-cycloadducts. In addition the multicomponent process only succeeded with AgSbF₆ derived chiral complex and very good yields and with identical enantioselections than that reported for the examples run with iminoesters as starters. Evaluating all

of these results, it was demonstrated how important resulted to be the anion in this sensitive enantioselective process, which is controlled by many parameters.

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- General procedure for the catalytic enantioselective 1,3-DC using silver salts:** A solution of the imino ester (1 mmol) and dipolarophile (1 mmol) in toluene (5 mL) was added to a suspension containing (*R*)- or (*S*)-Binap (0.05 mmol, 31 mg) and silver(I) salt (0.05 mmol) in toluene (5 mL). To the resulting suspension triethylamine (0.05 mmol, 7 μ L) was added and the mixture stirred at room temperature and in the absence of the light for 16-48 h (see main text). The reaction was filtered and the organic filtrate was directly evaporated and the residue was purified by recrystallization or by flash chromatography yielding pure *endo*-cycloadducts.
General procedure for the three-component catalytic enantioselective 1,3-DC using silver salts: To a suspension containing (*R*)- or (*S*)-Binap (0.05 mmol, 31 mg) and AgSbF₆ (0.05 mmol, 17 mg) in toluene (10 mL) was added the freshly distilled aldehyde (1 mmol), glycine methyl ester hydrochloride (1 mmol), the dipolarophile (1 mmol), and triethylamine (140 μ L, 1.05 mmol). The mixture was stirred at room temperature and in the absence of the light for 16-48 h (see main text). Following the same work-up than that described previously (above) final cycloadducts **3aa**, and **8c** were obtained.