

GMcal_TieLinesLL (v.2.3): Graphical User Interface (GUI) for the Topological Analysis of Calculated $G^{M(L)}$ Surfaces and Curves, including Tie-Lines, Hessian Matrix, Spinodal Curve, Critical Point Location, etc. for Binary and Ternary Liquid-Liquid Equilibrium (LLE) Data

(Topological Analysis of Liquid-Liquid Equilibrium Correlations)

Institutional Repository RUA: <http://hdl.handle.net/10045/51725>*

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USER INSTRUCTIONS*

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1. Download instructions

1. Download the file to your computer in a known folder: [GMcal_TieLinesLL.zip](#)
2. Unzip the file

2. Before using the GUI GMcal_TieLinesLL

1. In order to use this GUI it is necessary to have an excel file with an adequate structure, with the following information corresponding to the system under study. It is essential to respect the correct structure of the excel file. It is possible to use the different files exampleLL_test.xls included, as a draft for other similar case study or system (see **Appendix A: Structure of the excel data file**):
 - a. Values of $T^{(*)}$ and P for the corresponding LLE system.
 - b. $G^{\text{Excess}}(L)/RT$ Model selection variable: 1=NRTL, 2=UNIQUAC, 3 or higher=OTHER (that could be implemented by the user)
 - c. $G^E(L)/RT$ Model parameters:
 - For the **NRTL** model, the parameters used are: $\tau_{i,j}$ (dimensionless)= $A_{i,j}/(R \cdot T)$, $\alpha_{i,j}=\alpha_{j,i}$. i.e.: **9 parameters for a ternary system**: $\tau_{12}, \tau_{21}, \tau_{13}, \tau_{31}, \tau_{23}, \tau_{32}, \alpha_{12}, \alpha_{13}, \alpha_{23}$.
 - For the **UNIQUAC** model, the parameters used are: $\tau_{i,j}=\exp(-A_{i,j}/(R \cdot T))$, $r_i, q_i=s_i, q'_i, Z$. i.e.: **16 parameters for a ternary system**: $\tau_{12}, \tau_{21}, \tau_{13}, \tau_{31}, \tau_{23}, \tau_{32}, r_1, r_2, r_3, q_1, q_2, q_3, q'_1, q'_2, q'_3, Z$.
 - Classic NRTL or UNIQUAC models are the ones that can directly be used to calculate the GExcess. However, it is possible to use any other model by including in the MatLab function file: ModelGExcess.m the corresponding model (Model selection variable: **3 or higher= OTHER**). This alternative model could directly use a maximum of 19 parameters by using in the excel file the cells corresponding to the NRTL model plus the cells corresponding to the UNIQUAC model, cells, i.e.: E1:E6; H3:H5; K3:K5, L3:L6 and M3:M5. (see **Appendix A: Structure of the excel data file**). In the case of not using all the 19 parameters of the excel file, the

ones not used have to be equal to zero due to the fact that this function is called as:

function

```
y=ModelGExcess (ModelF, TEMP, x1, x2, x3, PARAM1, PARAM2, PARAM3, PARAM4, PARAM5, PARAM6, PARAM7, PARAM8, PARAM9, PARAM10, PARAM11, PARAM12, PARAM13, PARAM14, PARAM15, PARAM16, PARAM17, PARAM18, PARAM19).
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* As examples of the introduction of an additional model, three different examples are already implemented:

Model 3: a modified version of the NRTL model (e.g. used in Aspen Plus Software) for ternary systems with temperature dependent parameters ($\tau_{j,i} = a_{j,i} + b_{j,i}/T$ with 15 parameters: 6 $a_{i,j}$, 3 $\alpha_{i,j}$, 6 $b_{i,j}$).

Model 4: a modified version of the NRTL model (e.g. used in Aspen Plus Software) for binary systems with temperature dependent parameters (with 10 parameters).

Model 5: a modified version of the UNIQUAC model (e.g. used in Aspen Plus Software) for binary systems with temperature dependent parameters (with 13 parameters).

See Appendix A: Structure of the excel data file: **exampleLL_test3.xls**, **exampleLL_test4b-d.xls** and **exampleLL_test5.xls**, respectively.

* **In any case, just to remember that any model with no limitations in the number of parameters could be used by including the model and the specific parameters directly in the MatLab function file: ModelGExcess.m.**

* **Additionally, if the temperature of the system is constant, any dependence on the temperature of the temperature dependent parameters ($\tau_{j,i}$) could also be used by recalculating the constant values of $\tau_{j,i}$ at the given temperature.**

- d. The total number of experimental tie-lines (Nt) (max. value: 40). This value can be equal to **0** for ternary and binary LLE systems, to analyse only the corresponding parameters and model. In the case of binary systems, it is also necessary to write in cell B3 of the excel file the word **BINARY**^(*) (see the corresponding indications in sections 4.1 and 4.2 respectively).
- e. Experimental molar fractions
- f. Calculated molar fractions

➤ **Remark:** The proposed GUI GMcal_TieLinesLL can also be used to represent additionally the evolution of a **binary system** at different temperatures, including the calculation of the **critical solution temperature** (UCST, LCST or CML) if there exists (Ref.: *Ind. Eng. Chem. Res.* 2020, 59, 17, 8469-8479. <https://doi.org/10.1021/acs.iecr.0c00141>). In this case, an additional column including the temperature of each binary data (in increasing order) has also to be included with location = (M10:M50). Additionally, just in the case of the NRTL or UNIQUAC models, the parameters that have to be introduced are the $A_{j,i}(K)$ parameters (temperature independent variables).

A) **exampleLL_test4a.xls** (Figure 7a) using the NRTL model for a binary system (3 parameters)

B) **exampleLL_test4b(c and d).xls** (Figures 7b,c) using as “other model (max. 19 parameters directly from the excel file)”, the NRTL model for a binary system with temperature dependent parameters (10 parameters):

$$\tau_{ji} = a_{ji} + b_{ji}/T + e_{ji} \cdot \ln T + f_{ji} \cdot T$$

$$\alpha_{ji} = c_{ji} + d_{ji} (T - 273.15)$$

C) **exampleLL_test5.xls** using as “other model (max. 19 parameters directly from the excel file)”, the UNIQUAC model for a binary system with temperature dependent parameters (13 parameters):

$$\tau_{ji} = \exp(a_{ji} + b_{ji}/T + c_{ji} \cdot \ln T + d_{ji} \cdot T)$$

(For more information see: **Appendix A: Structure of the excel data file**)

3. **CLASIC GM MODELS RESUME**

➤ $G^{Mixture}/RT = G^{Ideal}/RT + G^{Excess}/RT = x_i \cdot \ln(x_i) + G^E/RT$

NRTL model:

$$\frac{G^E}{RT} = \sum_{i=1}^C x_i \cdot \frac{\sum_{j=1}^C \tau_{j,i} \cdot G_{j,i} \cdot x_j}{\sum_{k=1}^C G_{k,i} \cdot x_k}; \text{ with } \tau_{j,i} (\text{dimensionless}) = \frac{A_{j,i}}{RT}; G_{j,i} = \exp(-\alpha_{j,i} \cdot \tau_{j,i}); A_{i,i}=0$$

and $\alpha_{i,j} = \alpha_{j,i}$

UNIQUAC model:

$$\frac{G^E}{RT} = \frac{G_{Combinatorial}^E}{RT} + \frac{G_{Residual}^E}{RT} = \sum_{i=1}^C x_i \cdot \ln \frac{\phi_i}{x_i} + \frac{z}{2} \cdot \sum_{i=1}^C q_i \cdot x_i \cdot \ln \frac{\theta_i}{\phi_i} - \sum_{i=1}^C q'_i \cdot x_i \cdot \ln \left(\sum_{j=1}^C \theta'_j \cdot \tau_{j,i} \right)$$

$$\text{with: } \phi_i = \frac{r_i \cdot x_i}{\sum_{j=1}^C r_j \cdot x_j}; \theta_i = \frac{q_i \cdot x_i}{\sum_{j=1}^C q_j \cdot x_j}; \theta'_i = \frac{q'_i \cdot x_i}{\sum_{j=1}^C q'_j \cdot x_j}; \tau_{j,i} (\text{dimensionless}) = \exp\left(-\frac{A_{j,i}}{RT}\right)$$

z = Coordination number (usually with a value of 10)

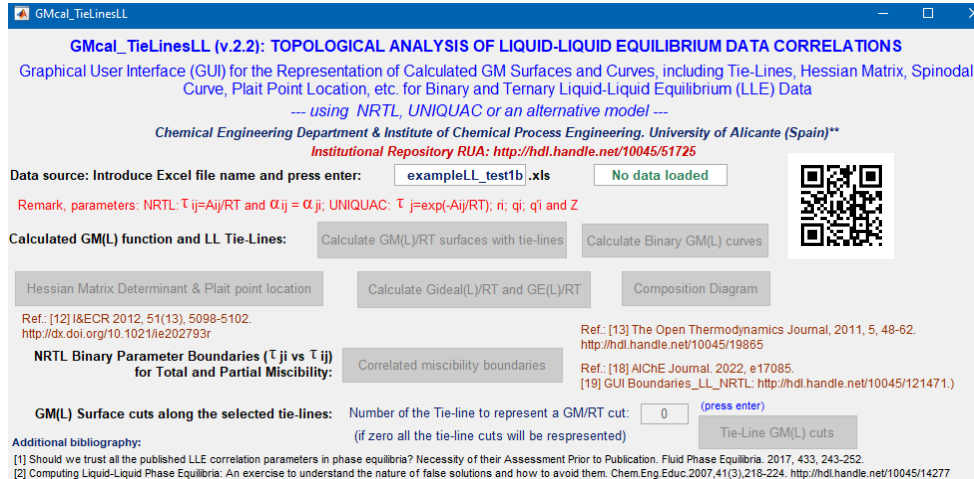
r_i = Volume parameter

q_i = Area parameter (or s_i)

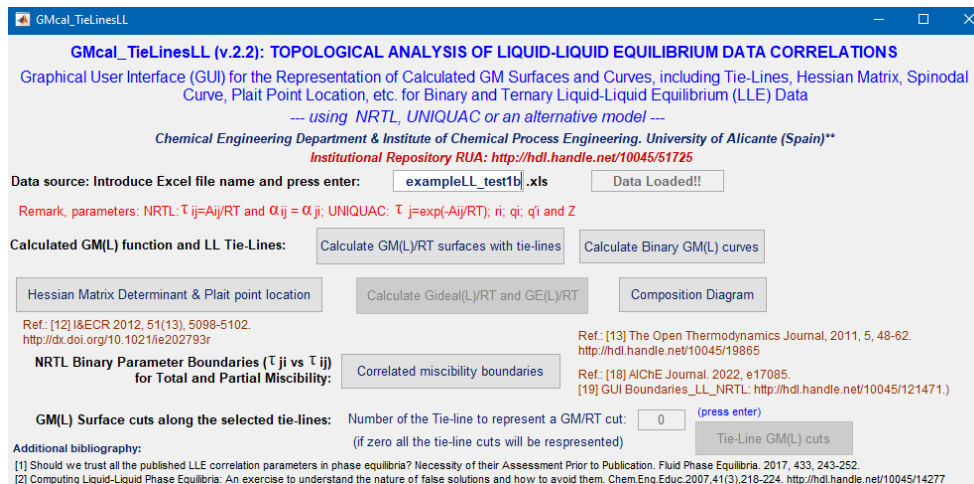
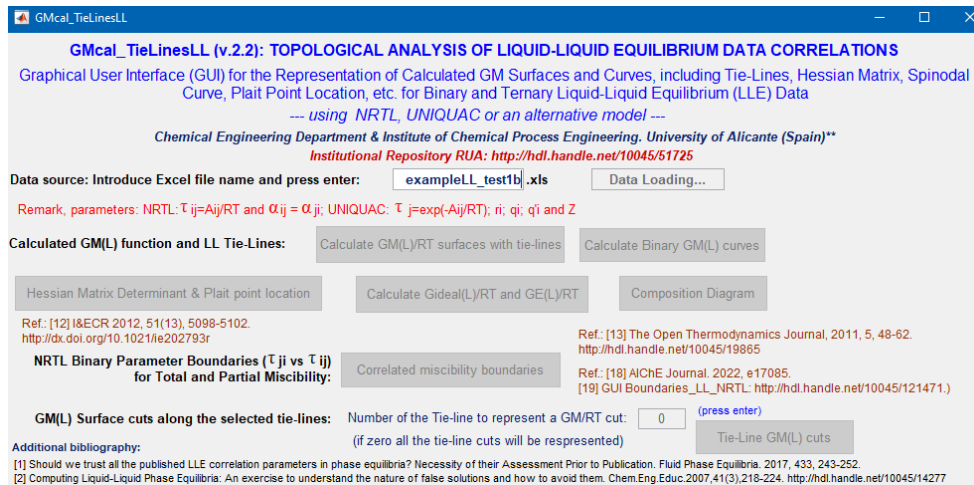
$q_i \neq q'_i$ for most of substances, except for water and some small alcohols.

4. **USING THE GUI GMcal TieLinesLL**

1. Open Matlab software
2. Once in MatLab, select the folder where the file **GMcal_TieLinesLL.zip** was unzipped as “current folder”.
3. Localize and execute the file **GMcal_TieLinesLL.p** from the MatLab Command Window (i.e.: writing **GMcal_TieLinesLL** in the Command Window, and pressing enter)



4. At this moment only the windows of the excel file name is active. Introduce the name of the excel data file corresponding to the system under study (e.g. exampleLL_test1b) and **press enter**. Please wait a moment for the data load process.



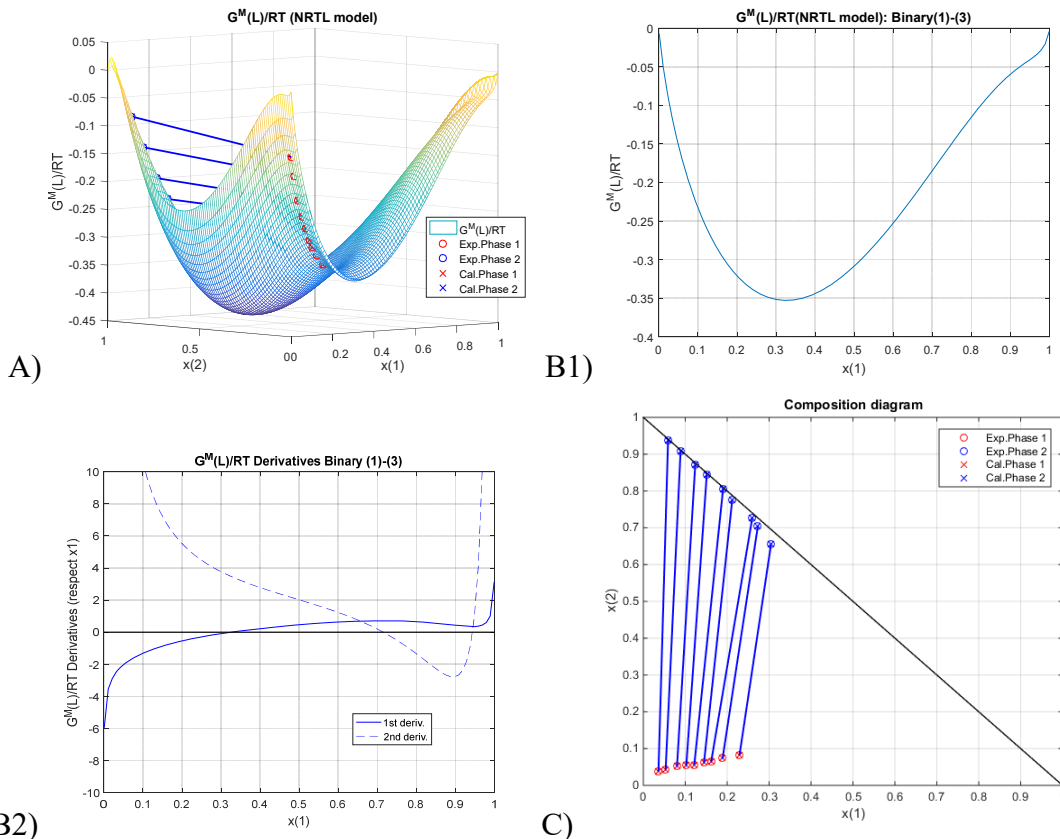
5. Now it is possible to create different diagrams by using the corresponding push buttons.

4.1) Ternary LLE systems at a constant temperature

- A) $G^{M(L)}/RT$ surface (calculated with the model and parameters defined), including also tie-lines (i.e. the corresponding experimental and calculated data included in the excel file loaded).
- B) Calculated Binary $G^{M(L)}$ curves, including 1st and 2nd derivatives (to facilitate the topological analysis).
- C) Composition diagram (in molar fractions).
- D) Hessian matrix determinant (σ), spinodal curve ($\sigma=0$), and Plait point visual location using additionally δ and δ^* matrix determinants, which have to be also equal to zero. (Ref.: *Industrial & Engineering Chemistry Research*. 2012, 51 (13), 5098-5102. <http://dx.doi.org/10.1021/ie202793r>):

$$\sigma = \begin{vmatrix} \frac{\partial^2 G^M}{\partial x_1^2} & \frac{\partial^2 G^M}{\partial x_1 \partial x_2} \\ \frac{\partial^2 G^M}{\partial x_2 \partial x_1} & \frac{\partial^2 G^M}{\partial x_2^2} \end{vmatrix} = 0; \delta = \begin{vmatrix} \frac{\partial \sigma}{\partial x_1} & \frac{\partial \sigma}{\partial x_2} \\ \frac{\partial^2 G^M}{\partial x_2 \partial x_1} & \frac{\partial^2 G^M}{\partial x_2^2} \end{vmatrix} = 0; \delta^* = \begin{vmatrix} \frac{\partial \sigma}{\partial x_2} & \frac{\partial \sigma}{\partial x_1} \\ \frac{\partial^2 G^M}{\partial x_1 \partial x_2} & \frac{\partial^2 G^M}{\partial x_1^2} \end{vmatrix} = 0$$

- E) Correlated NRTL binary Miscibility Boundaries (e.g. for Total and Partial Miscibility): $\tau_{j,i}$ vs $\tau_{i,j}$ (Ref.: *The Open Thermodynamics Journal*, 2011, 5, 48-62: <http://dx.doi.org/10.2174/1874396X01105010048> (for $\alpha_{ij}=0.2$) and *AIChE* 2022, e17805: <https://dx.doi.org/10.1002/aic.17805-RUA>, 2022: <http://hdl.handle.net/10045/121471>, for α_{ij} between 0 and 0.95). The $\tau_{i,j}$ binary parameters (included in the excel file) are also represented, to check their adequate location.



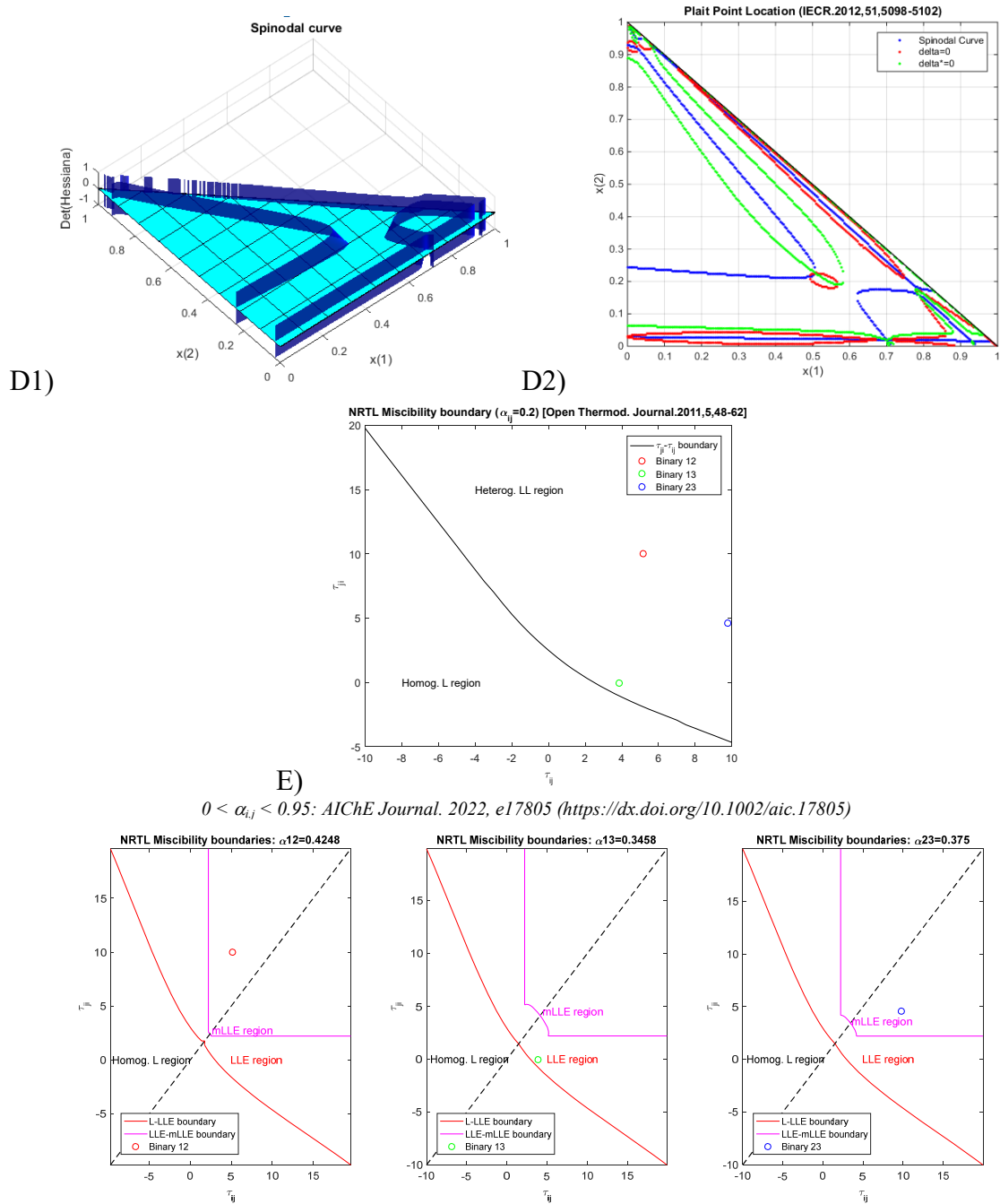


Figure 1. Different diagrams corresponding to exampleLL_test1a.xls (LLE ternary system). In this example, the NRTL parameters used are **NOT COHERENT** with the experimental LL data.

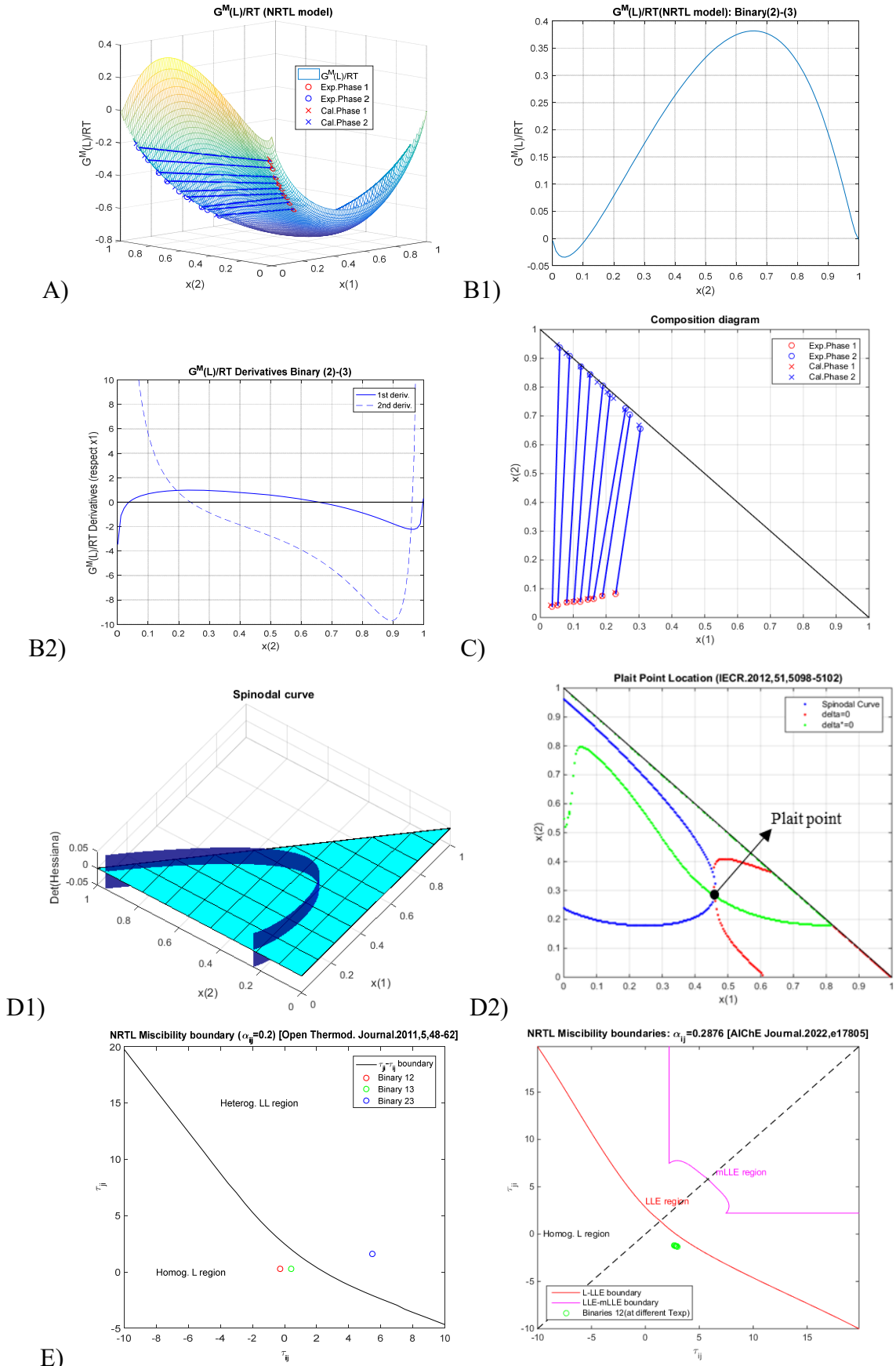
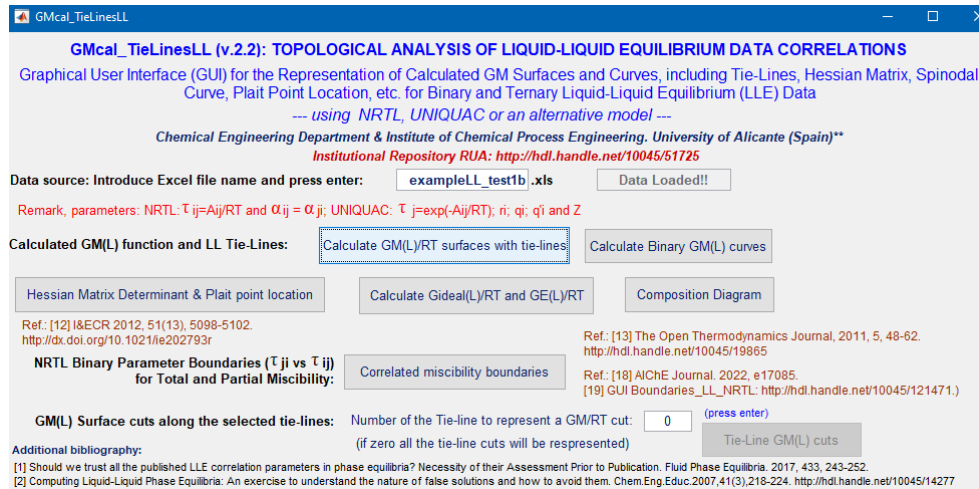


Figure 2. Different diagrams corresponding to exampleLL_test1b.xls (LLE ternary system). In this example, the NRTL parameters used are **COHERENT** with the experimental LL data.

6. Once the $G^M(L)/RT$ surface (A) is calculated, two more options are active:



- F) $G^{\text{Ideal}}(L)/RT$ and $G^E(L)/RT$ surfaces (ideal and excess contributions of the $G^M(L)/RT$ function calculated with the model parameters included in the excel data file): Push button.
- G) $G^M(L)/RT$ surface cuts along the selected Experimental Tie-Lines. In this case, the ordinal number corresponding to a tie-line (from 0 to N_t) has to be introduced (the value 0 will select the representation of the different cuts along all the experimental and calculated tie-lines). Then **press enter**. After that, push the button “Tie Line $GM(L)$ ” to represent the $G^M(L)/RT$ surface cut is active and can be used.

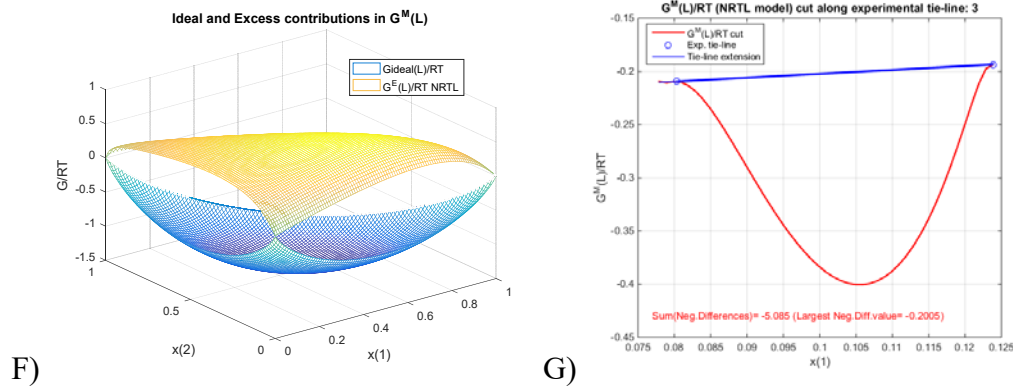


Figure 3. Different diagrams corresponding to exampleLL_test1a.xls: Ideal and Excess contribution, and $G^M(L)/RT$ cut along a tie-line. In this example, the NRTL parameters used are **NOT COHERENT**.

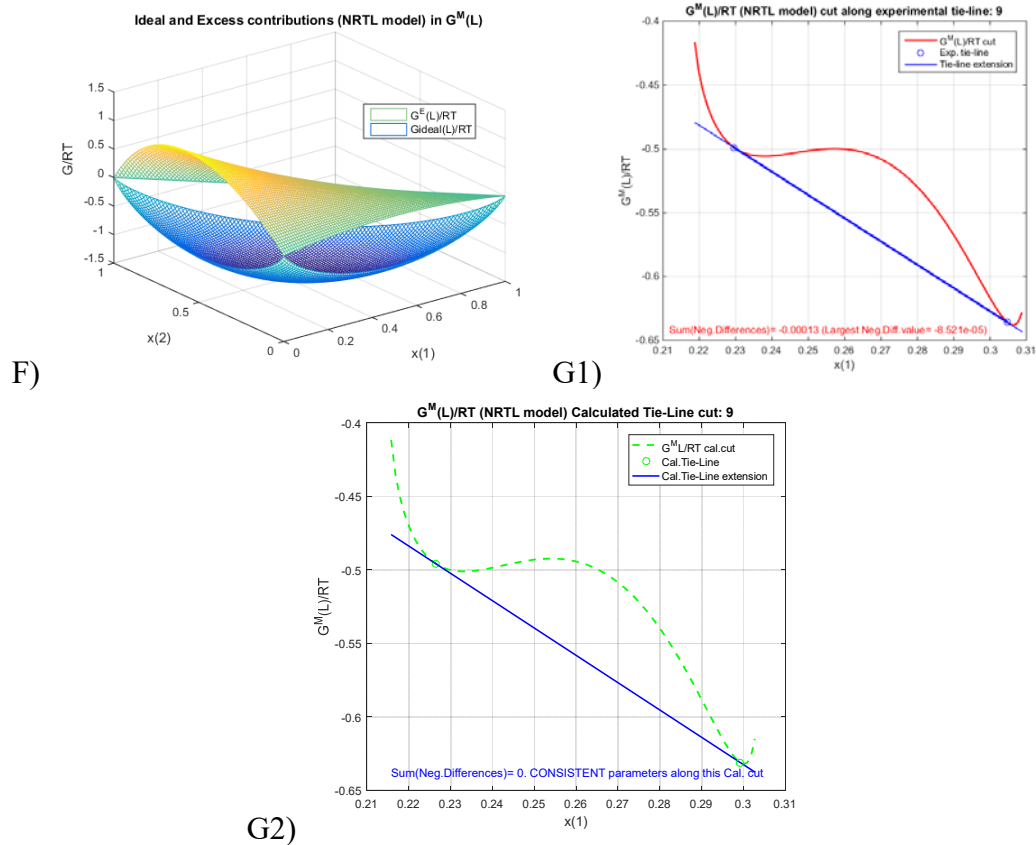


Figure 4. Different diagrams corresponding to exampleLL_test1b.xls: Ideal and Excess contribution, and $G^M(L)/RT$ cut along an exp. and cal. tie-line. In this example, the NRTL parameters used are **COHERENT**.

When the number of the Tie Line is different from 0, two additional figures are also generated in ternary systems to facilitate and complete the visual analysis (Figures 5). On one hand, a new 3D figure representing the complete $G^M(L)/RT$ surface (in modified units for drawing horizontal the tangent planes for the selected tie line, to facilitate the viewing and the coherence of the results). This figure includes the selected tie-line and the tangent planes at both extremes of the corresponding calculated or experimental (Figures 5.a1 and 5.b1). The second figure represents additionally the difference between $(G^M(L)/RT)_{\text{Surface(model)}}$ and $(G^M(L)/RT)_{\text{tangent plane}}$ in the whole range of compositions, including the way along the tie line cut and the tangent planes at its both extremes (see Figures 5.a2 and 5.b2). Evidently, this difference in the equilibrium compositions should be equal to zero.

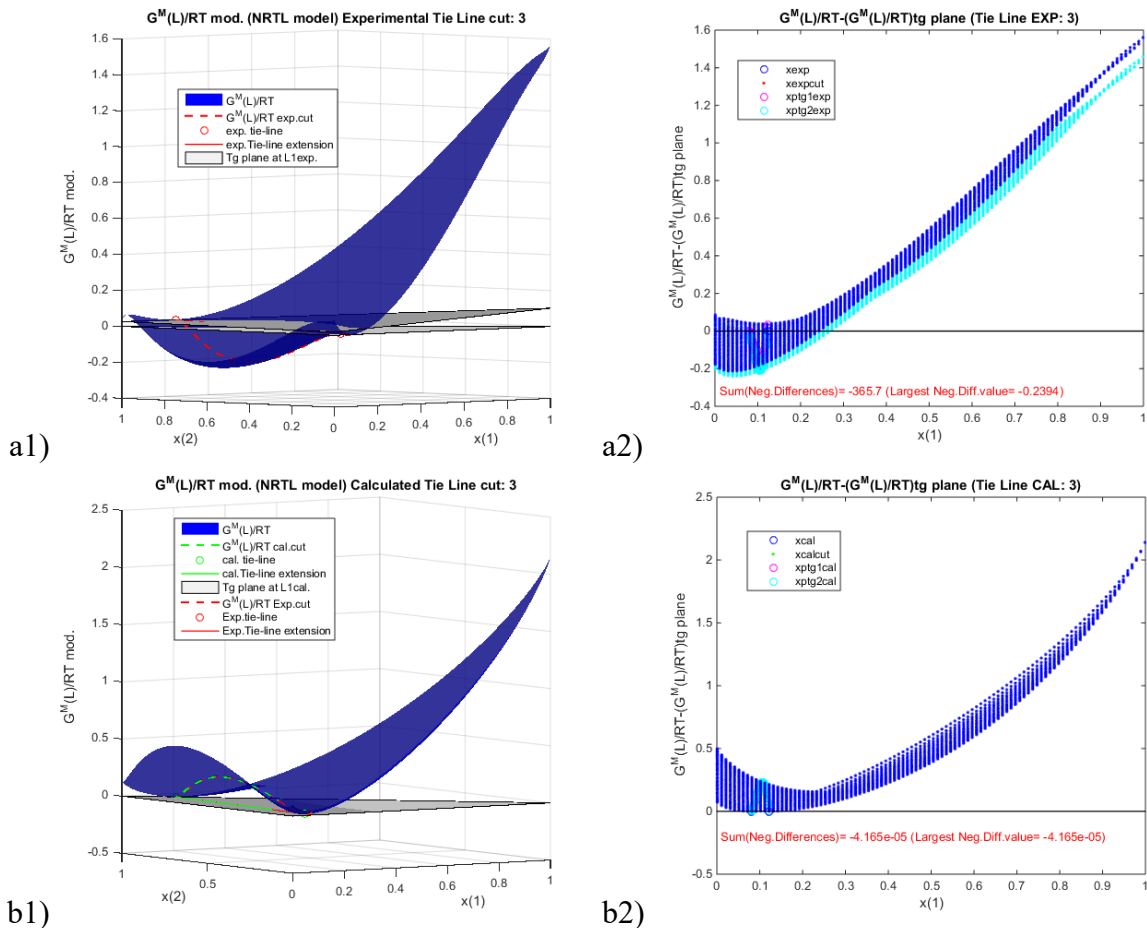
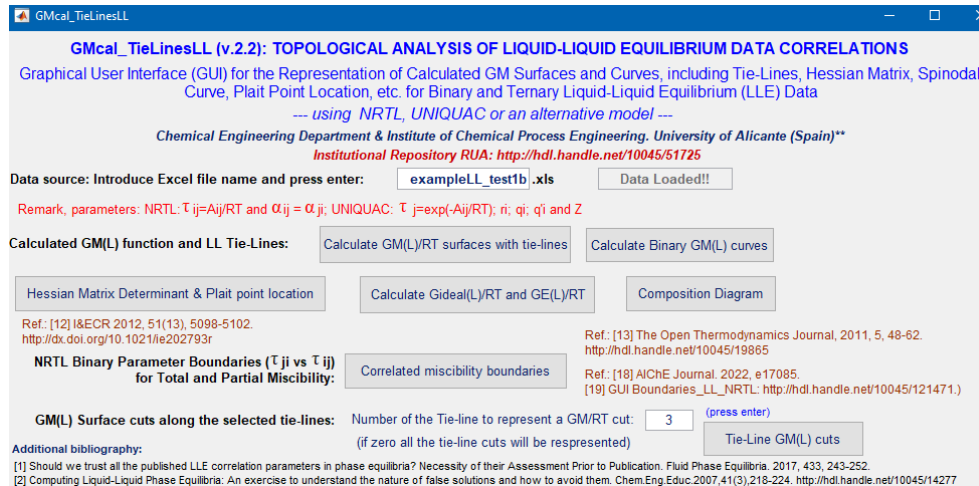


Figure 5. Complete $G^M(L)/RT$ surface including the selected tie-line and the tangent planes at both extremes of the tie-line. Remark: the values of the $G^M(L)/RT$ surface have been modified to draw horizontal tangent planes to facilitate the viewing. a) Example with **NOT COHERENT** NRTL parameters, b) Example with **COHERENT** parameters.

In order to complement this visual analysis in a rigorous way, from the version v.2.2. of this GUI, the summation of all the negative differences $[(G^M(L)/RT)_{Surface(model)} - (G^M(L)/RT)_{tangent\ plane}]$, and the largest negative difference, for each tie line (and the global for all the tie lines) are also calculated and showed in the Command Windows of MatLab (Figure 6). This calculation is

carried out in the cut along the tie lines and additionally in ternary systems, in the whole range of composition. Obviously, these summations should be zero for totally consistent parameters with the expected experimental behaviour.

```

Command Window

Cut along exp. Tie Line 1: Sum(Neg.Differences)= 0 (Largest Neg.Diff.value= 0)
CONSISTENT parameters along this exp. cut

Cut along cal. Tie Line 1: Sum(Neg.Differences)= 0 (Largest Neg.Diff.value= 0)
CONSISTENT parameters along this cal. cut
>>> Tangent Plane at Tie Line cal.1: Sum(Neg.Differences)= -0.00097828 (Largest(Neg.Diff.value)= -0.00071431)

Cut along exp. Tie Line 2: Sum(Neg.Differences)= 0 (Largest Neg.Diff.value= 0)
CONSISTENT parameters along this exp. cut

Cut along cal. Tie Line 2: Sum(Neg.Differences)= 0 (Largest Neg.Diff.value= 0)
CONSISTENT parameters along this cal. cut
>>> Tangent Plane at Tie Line cal.2: Sum(Neg.Differences)= -0.0014636 (Largest(Neg.Diff.value)= -0.00073019)

...

Cut along exp. Tie Line 8: Sum(Neg.Differences)= -8.382e-05 (Largest Neg.Diff.value= -8.382e-05)

Cut along cal. Tie Line 8: Sum(Neg.Differences)= 0 (Largest Neg.Diff.value= 0)
CONSISTENT parameters along this cal. cut
>>> Tangent Plane at Tie Line cal.8: Sum(Neg.Differences)= -0.0035389 (Largest(Neg.Diff.value)= -0.00060293)

Cut along exp. Tie Line 9: Sum(Neg.Differences)= -0.00012998 (Largest Neg.Diff.value= -8.5209e-05)

Cut along cal. Tie Line 9: Sum(Neg.Differences)= 0 (Largest Neg.Diff.value= 0)
CONSISTENT parameters along this cal. cut
>>> Tangent Plane at Tie Line cal.9: Sum(Neg.Differences)= -0.00089197 (Largest(Neg.Diff.value)= -0.00023245)

--> TOTAL Sum(Neg.Differences) along all exp. cuts= -0.005024 (Largest Neg.Diff.value= -0.0017592)

----> TOTAL Sum(Neg.Differences) along all cal. cuts= 0 (Largest Neg.Diff.value= 0)
CONSISTENT PARAMETERS ALONG ALL CAL. CUTS
>>>> TOTAL Sum(Neg.Differences) along all Tangent Planes at cal. Tie Lines= -0.18865 (Largest Neg.Diff.value= -0.0047468)
fx >>
    
```

Figure 6. Example of the additional information that appears at the Command Window after pushing the "Tie-Line GM(L) cuts" button.

As commented before, this GMcal_TieLinesLL GUI can be also used without any experimental data, just to analyse only the corresponding parameters and model. In the case of ternary LLE systems, this can be done only by indicating that the total number of experimental tie-lines (N_t) is equal to 0 (cell B4 in the excel file). E.g. **exampleLL_test0T.xls**.

4.2) Binary LLE systems at different temperatures

As commented before, the proposed GMcal_TieLinesLL GUI can also be used to represent additionally the evolution of a **binary system** at different temperatures, including the calculation of the **critical solution temperature** (UCST, LCST or CML) in a predefined range of temperatures, if there exists (Ref.: *Ind. Eng. Chem. Res.* 2020, 59, 17, 8469–8479. <https://doi.org/10.1021/acs.iecr.0c00141>). e.g. Figure 7 and **exampleLL_test4a-4d.xls**.

In this case, it is necessary to write in cell B3 of the excel file the word BINARY. Additionally, the components used have to be 1 and 2, and an additional column including the temperature of each binary data (in increasing order) has also to be included with location=(M10:M50). And also, in the case of the NRTL or UNIQUAC models, the parameters that have to be introduced are the A_{ji} (K) parameters (temperature independent variables). See **Appendix A: Structure of the excel data file**.

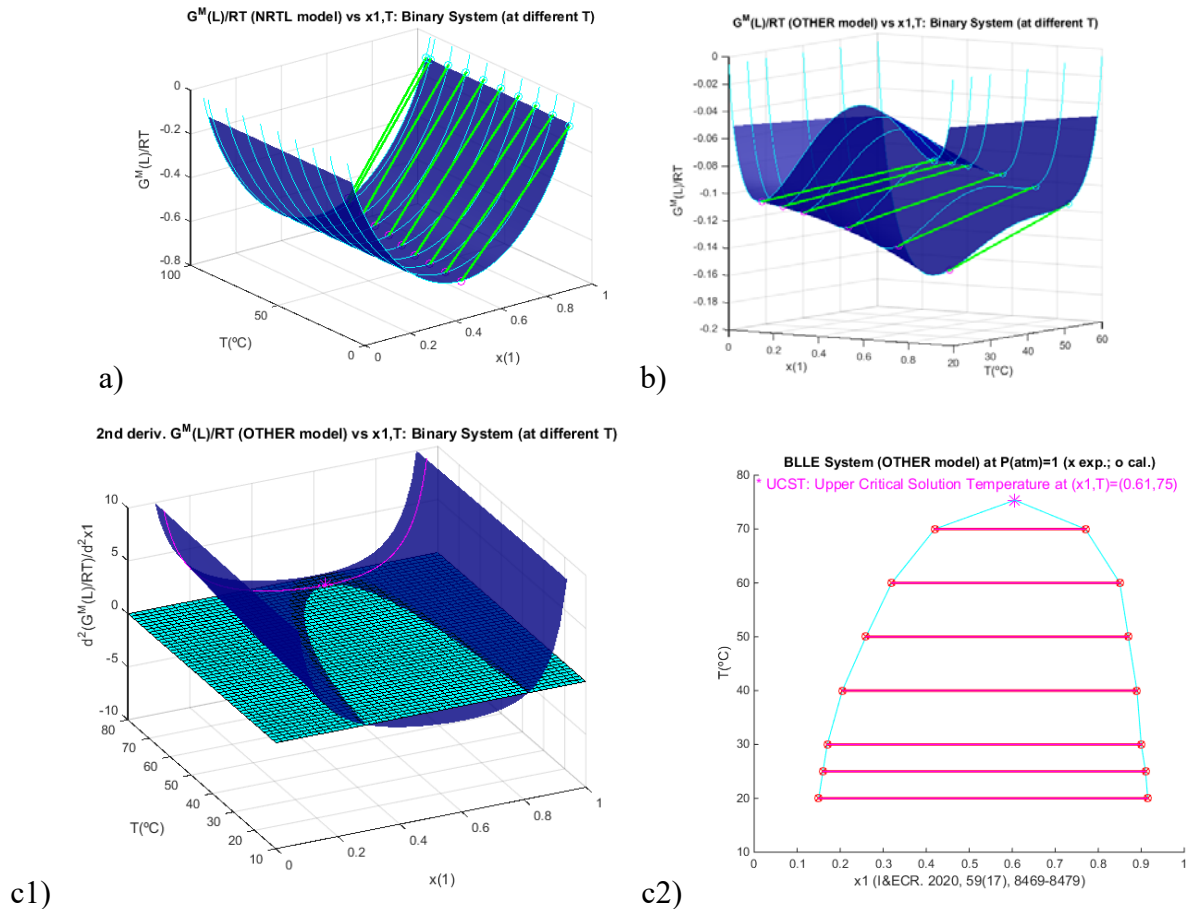


Figure 7. Additional graphics for LLE binary systems at different temperatures: Evolution of the $G^{M(L)}/RT$ surface with the composition and temperature (including experimental tie-lines). a) Example with **not coherent** NRTL parameters (e.g. exampleLL_test4a.xls). b) Example with **coherent** parameters (e.g. exampleLL_test4b.xls). c) Determination of the critical solution temperature [18] (e.g. exampleLL_test4b.xls).

In the case of using this GUI GMcal_TieLinesLL for binary systems without any experimental data just to analyse only the corresponding parameters and model, it is necessary:

- to write in the cell B3 of the excel file the word **BINARY**, and also
- to indicate that the total number of experimental tie-lines (Nt) is equal to 0 (cell B4 in the excel file).
- It is possible additionally to establish a range of temperatures to look for Critical Solution Temperatures. To do that can be used cells M10 and M11 (in increasing order). The range of temperatures defined by defect is 0-150°C.

E.g. exampleLL_test0B.xls.

Finally, a new excel file for other binary or ternary case of study can be loaded to start again, or exit to the applications by using the exit push button.

For further information or comments. Corresponding author: ja.labarta@ua.es
(ORCID  <http://orcid.org/0000-0002-4870-2031>)

After using this GUI GMcal_TieLinesLL, please complete the following quick inquiry to know your opinion about the graphical user interface developed:



English version. general user:

https://docs.google.com/forms/d/e/1FAIpQLSc_N4Oq0EMCkRjRgA1KeAwG8R8ZaxfZQ4lw-H3EjkMfruJvsw/viewform?c=0&w=1&usp=mail_form_link



Spanish version. Specific for Spanish students

<https://docs.google.com/forms/d/e/1FAIpQLSfYkFzPrWK9iMummo210phzd3sgfEI-M-4U7TPIywc8ZpEmLw/viewform?c=0&w=1>



Facebook:

<http://www.facebook.com/GUI-for-the-analysis-of-the-Gibbs-stability-criteria-1861744990767271/>

REMARK: IF YOU WANT TO RECEIVE A NOTIFICATION WITH POSSIBLE UPDATES OF THIS GUI. PLEASE SENT AN E-MAIL TO: ja.labarta@ua.es

APPENDIX A: STRUCTURE OF THE EXCEL DATA FILE FOR THE GMcal_TieLinesLL GUI

It is essential to respect the correct structure of the excel file. It is possible to use the file exampleLL_test.xls included as a draft for other case studies or systems.

- 1.- The active sheet of the excel file has to call it: **Example**
- 2.- The different data needed for the calculations and graphical representations has to be located in the following concrete cells of the active excel sheet:

Variable	Concrete cell location in the active (loaded) excel sheet
Pressure (atm)	B2
T(°C) ^(*)	B3
Total number of experimental tie-lines (Nt≤40) ^(**)	B4
Model Used: (1: NRTL; 2: UNIQUAC; 3 or higher: OTHER MODEL (that has to be implemented previously in the ModelGExcess.m file)	B5
For NRTL model	---
NRTL binary parameter $T_{12}=A_{12}/RT$	E1 (** PARAM1)
NRTL binary parameter $T_{21}=A_{21}/RT$	E2 (** PARAM2)
NRTL binary parameter $T_{13}=A_{13}/RT$	E3 (** PARAM3)
NRTL binary parameter $T_{31}=A_{31}/RT$	E4 (** PARAM4)
NRTL binary parameter $T_{23}=A_{23}/RT$	E5 (** PARAM5)
NRTL binary parameter $T_{32}=A_{32}/RT$	E6 (** PARAM6)
NRTL binary parameter $\alpha_{12}=\alpha_{21}$	H3 (** PARAM7)
NRTL binary parameter $\alpha_{13}=\alpha_{31}$	H4 (** PARAM8)
NRTL binary parameter $\alpha_{23}=\alpha_{32}$	H5 (** PARAM9)
For UNIQUAC model	---
$T_{12}=\exp(-A_{12}/RT)$	E1
$T_{21}=\exp(-A_{21}/RT)$	E2
$T_{13}=\exp(-A_{13}/RT)$	E3
$T_{31}=\exp(-A_{31}/RT)$	E4
$T_{23}=\exp(-A_{23}/RT)$	E5
$T_{32}=\exp(-A_{32}/RT)$	E6
r ₁	K3 (** PARAM10)
r ₂	K4 (** PARAM11)
r ₃	K5 (** PARAM12)
q ₁	L3 (** PARAM13)
q ₂	L4 (** PARAM14)
q ₃	L5 (** PARAM15)
q' ₁	M3 (** PARAM16)
q' ₂	M4 (** PARAM17)
q' ₃	M5 (** PARAM18)
Z	L6 (** PARAM19)

EXPERIMENTAL TIE-LINES*	
Molar fraction component 1 in Phase 1	A10:A50
Molar fraction component 2 in Phase 1	B10:B50
Molar fraction component 3 in Phase 1	C10:C50
Molar fraction component 1 in Phase 2	D10:D50
Molar fraction component 2 in Phase 2	E10:E50
Molar fraction component 3 in Phase 2	F10:F50
CALCULATED TIE-LINES✓	
Molar fraction component 1 in Phase 1	G10:G50
Molar fraction component 2 in Phase 1	H10:H50
Molar fraction component 3 in Phase 1	I10:I50
Molar fraction component 1 in Phase 2	J10:J50
Molar fraction component 2 in Phase 2	K10:K50
Molar fraction component 3 in Phase 2	L10:L50

(**) Initially, the maximum value of the total number of tie-lines admitted is: **40**. This value can be equal to **0** for ternary and binary LLE systems, to analyze only the corresponding parameters and model. In the case of binary systems, it is also necessary to write in cell B3(*) of the excel file the word **BINARY** (see the corresponding examples: exampleLL_test0T.xls and exampleLL_test0B.xls).

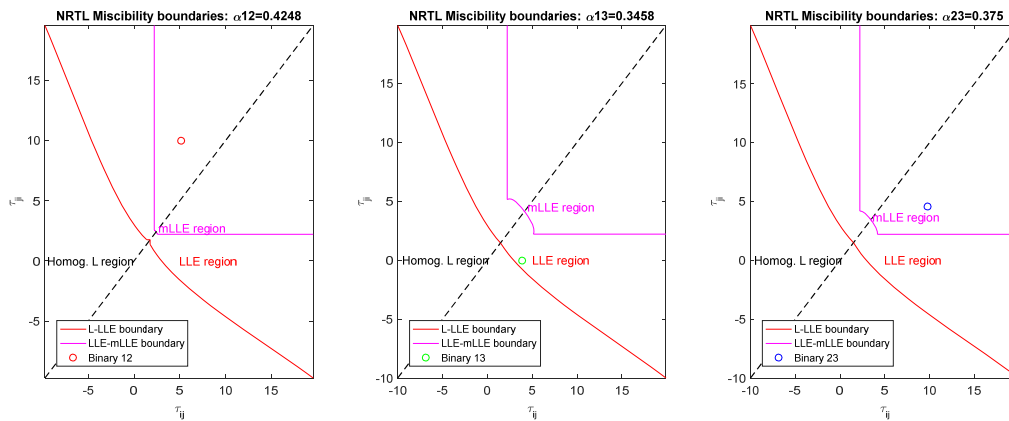
✓ If calculated tie-lines are not available, the corresponding cells have to be empty.

List of test files included (exampleLL test.xls)

LLE TERNARY SYSTEMS

➤ exampleLL_test1a.xls: LLE ternary system. NRTL model (without calculated data).

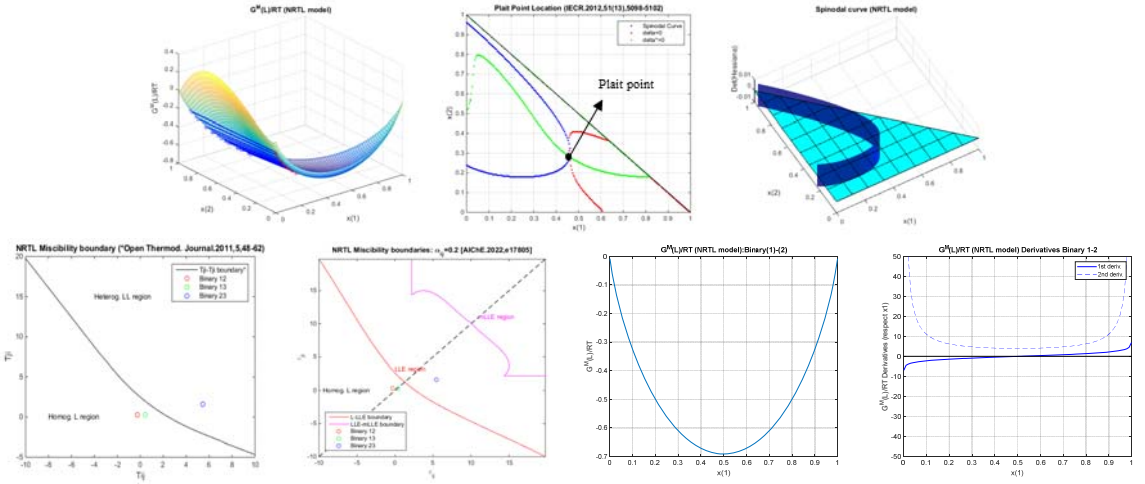
	A	B	C	D	E	F	G	H	I	J	K	L	M
1	Parameters			T12	5,149	Tij=Aij/(R·T)							
2	P(atm)=	1		T21	9,999		NRTL						
3	T° (°C)=	25		T13	3,868		ALFA12=	0,42					
4	N.Tie-Lines=	9		T31	-0,068		ALFA13=	0,35					
5	Model used:	1	1: NRTL	T23	9,771		ALFA23=	0,38					
6			2: UNIQUAC	T32	4,578								
7	Experimental Data		3: OTHER				Calculated Data						
8	fase 1			fase 2			fase 1			fase 2			
9	x1	x2	x3	x1	x2	x3	x1	x2	x3	x1	x2	x3	
10	0,036	0,039	0,925	0,059	0,938	0,003							
11	0,053	0,043	0,904	0,089	0,908	0,003							
12	0,080	0,052	0,867	0,124	0,871	0,005							
13	0,102	0,054	0,843	0,151	0,843	0,005							
14	0,122	0,056	0,823	0,190	0,805	0,005							
15	0,145	0,062	0,794	0,212	0,775	0,013							
16	0,162	0,064	0,774	0,260	0,726	0,014							
17	0,189	0,074	0,737	0,273	0,704	0,023							
18	0,230	0,083	0,687	0,305	0,656	0,039							

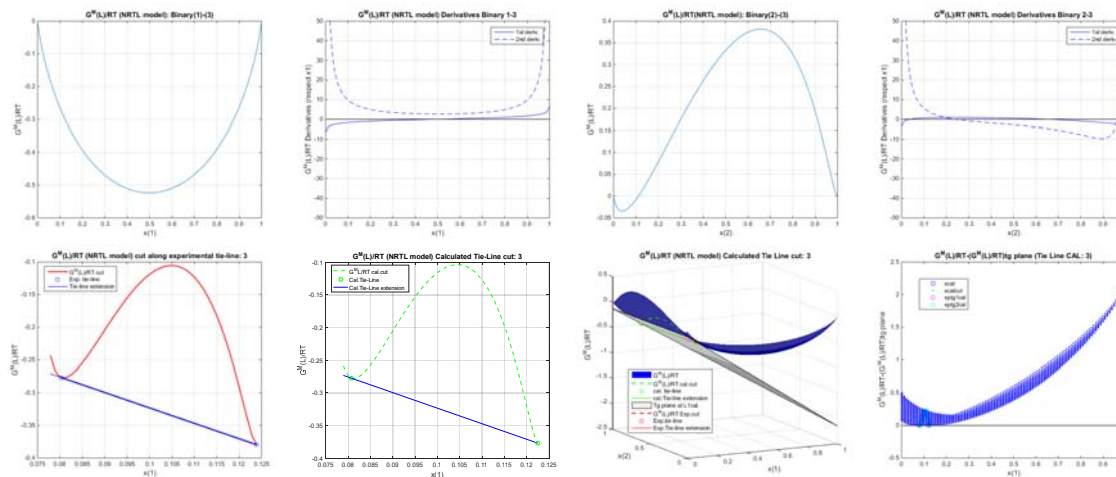


$0 < \alpha_{i,j} < 0.95$: *AIChE Journal*, 2022, e17805 (<https://dx.doi.org/10.1002/aic.17805>)

➤ **exampleLL_test1b.xls: LLE ternary system. NRTL model (with calculated data).**

Parameters		T ₁₂		T ₁₃		T ₂₃		T _{ij} =A _{ij} /T		NRTL		Fluid Phase Equilibria, 2017, 433, 243-252.					
1	P(α _m)=	1	T21	0.2934996									https://doi.org/10.1016/j.fluid.2016.11.009				
2	T(β _m)=	25	T13	0.4728198									https://doi.org/10.1016/j.fluid.2016.11.009				
3	N Tie-Lines=	9	T31	0.2011906									https://doi.org/10.1016/j.fluid.2016.11.009				
4	Model used=	1	1: NRTL	T23	5.4843847								https://doi.org/10.1016/j.fluid.2016.11.009				
5	Tiebreak type=	1	2: UNQUAC	T32	1.6203576								https://doi.org/10.1016/j.fluid.2016.11.009				
6	Experimental Data	3 or less	Calculated Data		Case 1		Case 2		Case 3		Mean dev. between cal. and exp. conc. in mol fractions						
7	Case 1		x1	x2	x3	x1	x2	x3	x1	x2	x3	0.00052896	0.00052896	0	0.001865276	0.001865276	0.000222888
8	Case 2		x1	x2	x3	x1	x2	x3	x1	x2	x3	0.00466184	0.00466184	0	0.00052489	0.00052489	0.000595404
9	Case 3		x1	x2	x3	x1	x2	x3	x1	x2	x3	0.00384462	0.00384462	0	0.0118541	0.0118541	0.00049582
10	Case 4		x1	x2	x3	x1	x2	x3	x1	x2	x3	0.000422474	0.000422474	0	0.000525277	0.000525277	0.000879245
11	Case 5		x1	x2	x3	x1	x2	x3	x1	x2	x3	0.000433064	0.000433064	0	0.003272568	0.003272568	0.000288001
12	Case 6		x1	x2	x3	x1	x2	x3	x1	x2	x3	0.000202347	0.000202347	0	0.0147432	0.0147432	0.004891845
13	Case 7		x1	x2	x3	x1	x2	x3	x1	x2	x3	0.00342791	0.00342791	0	0.00944075	0.00944075	0.0005154
14	Case 8		x1	x2	x3	x1	x2	x3	x1	x2	x3	0.003232965	0.003232965	0	0.03787969	0.03787969	0.00270041
15	Case 9		x1	x2	x3	x1	x2	x3	x1	x2	x3	0.00295732	0.00295732	0	0.01628102	0.01628102	0.000780562
16	Case 10		x1	x2	x3	x1	x2	x3	x1	x2	x3	0.003053532	0.003053532	0	0.00552728	0.00552728	0.000530377
17	Case 11		x1	x2	x3	x1	x2	x3	x1	x2	x3						
18	Case 12		x1	x2	x3	x1	x2	x3	x1	x2	x3						
19	Case 13		x1	x2	x3	x1	x2	x3	x1	x2	x3						
20	Case 14		x1	x2	x3	x1	x2	x3	x1	x2	x3						
21	Case 15		x1	x2	x3	x1	x2	x3	x1	x2	x3						
22	Case 16		x1	x2	x3	x1	x2	x3	x1	x2	x3						
23	Case 17		x1	x2	x3	x1	x2	x3	x1	x2	x3						
24	Case 18		x1	x2	x3	x1	x2	x3	x1	x2	x3						
25	Case 19		x1	x2	x3	x1	x2	x3	x1	x2	x3						
26	Case 20		x1	x2	x3	x1	x2	x3	x1	x2	x3						
27	Case 21		x1	x2	x3	x1	x2	x3	x1	x2	x3						
28	Case 22		x1	x2	x3	x1	x2	x3	x1	x2	x3						
29	Case 23		x1	x2	x3	x1	x2	x3	x1	x2	x3						
30	Case 24		x1	x2	x3	x1	x2	x3	x1	x2	x3						
31	Case 25		x1	x2	x3	x1	x2	x3	x1	x2	x3						
32	Case 26		x1	x2	x3	x1	x2	x3	x1	x2	x3						
33	Case 27		x1	x2	x3	x1	x2	x3	x1	x2	x3						
34	Case 28		x1	x2	x3	x1	x2	x3	x1	x2	x3						
35	Case 29		x1	x2	x3	x1	x2	x3	x1	x2	x3						
36	Case 30		x1	x2	x3	x1	x2	x3	x1	x2	x3						
37	Case 31		x1	x2	x3	x1	x2	x3	x1	x2	x3						
38	Case 32		x1	x2	x3	x1	x2	x3	x1	x2	x3						





➤ e.g. **exampleLL_test2.xls**: LLE ternary system. UNIQUAC model (without calculated data).

	A	B	C	D	E	F	G	H	I	J	K	L	M	N	
1	Parameters			T12	1,43844874	Tij=exp(-Aij/T)				Remark: qi and q'i are different only for water and some small alcohols					
2	P(atm)=	1		T21	2,16509148					UNIQUAC		qi	q'i		
3	T (°C)=	25		T13	0,40101354					1	0,92	1,4	1,4		
4	N Tie-Lines=	5		T31	0,10300668					2	2,1055	1,972	1,972		
5	Model used:	2	1: NRTL	T23	1,12206826					3	3,9228	2,968	2,968		
6			2: UNIQUAC	T32	0,69486078							z	10		
7	Experimental Data			3: OTHER			Calculated Data								
8	fase 1			fase 2			fase 1			fase 2					
9	x1	x2	x3	x1	x2	x3	x1	x2	x3	x1	x2	x3			
10	0,9997	0,0000	0,0003	0,0102	0,0000	0,9898									
11	0,8303	0,1667	0,0030	0,0151	0,0389	0,9461									
12	0,7351	0,2570	0,0079	0,0224	0,0860	0,8916									
13	0,6141	0,3644	0,0215	0,0382	0,1716	0,7902									
14	0,4754	0,4684	0,0561	0,0689	0,2963	0,6348									

➤ **exampleLL_test3.xls**: LLE ternary system. Alternative model (NRTL with $\tau_{ij} = a_{ij} + b_{ij}/T$). 15 parameters: 6 a_{ij} , 3 α_{ij} , 6 b_{ij} . Without calculated data.

As commented before, in an alternative way, cells E1:E6; H3:H5; K3:K5; L3:L5; M3:M5 and L6 (in this order) could be used to introduce a maximum of 19 parameters directly from the excel file for any OTHER model (different to NRTL or UNIQUAC) to calculate the Gibbs Excess Function using the corresponding matlab function file (named: **ModelGExcess.m**) and keeping the following structure, variables and name: **Function**
`y=ModelGExcess(modelF, TEMP, x1, x2, x3, PARAM1, PARAM2, PARAM3, PARAM4, PARAM5, PARAM6, PARAM7, PARAM8, PARAM9, PARAM10, PARAM11, PARAM12, PARAM13, PARAM14, PARAM15, PARAM16, PARAM17, PARAM18, PARAM19)`

In any case, just to remember that if the temperature of the system is constant, any dependence of possible temperature dependent parameters (τ_{ij}) could also be used by recalculating the constant values of τ_{ij} at the given temperature. Additionally, also remark that any model with no limitations in the number of parameters could be use by including the model and the specific parameters directly in the MatLab function file: **ModelGExcess.m.**

As example of the introduction of an additional model, in the **exampleLL_test3.xls** test file, a modified version of the NRTL model (e.g. used in Aspen Plus Software) for ternary systems with temperature dependent parameters: $\tau_{ji} = a_{ji} + b_{ji}/T$ (15 parameters: 6 a_{ij} , 3 α_{ij} , 6 b_{ij}) has also been introduced (this "other model" has been numbered in the **ModelGExcess.m** file as **Model 3**), following the previous indications where:

a12	E1 (** PARAM1)
a21	E2 (** PARAM2)
a13	E3 (** PARAM3)
a31	E4 (** PARAM4)
a23	E5 (** PARAM5)
a32	E6 (** PARAM6)
-----	-----
$\alpha_{12}=\alpha_{21}$	H3 (** PARAM7)
$\alpha_{13}=\alpha_{31}$	H4 (** PARAM8)
-----	-----
$\alpha_{23}=\alpha_{32}$	H5 (** PARAM9)
b12	K3 (** PARAM10)
b21	K4 (** PARAM11)
b13	K5 (** PARAM12)
b31	L3 (** PARAM13)
b23	L4 (** PARAM14)
b32	L5 (** PARAM15)

A	B	C	D	E	F	G	H	I	J	K	L	M	N	O	P	Q	R
Parameters													Fluid Phase Equilibria. 2017, 433, 243-252.				
P(atm)=	1		param.1	0,281		more parameters 7-9			more parameters 10-19				http://dx.doi.org/10.1016/j.fluid.2016.11.009				
T* (°C)=	20		param.2	-2,110		param.7	0,20		param.10/13/16:	-23,551	-938,986	0	GUI v.2.0 for the representation of GM surfaces				
N.Tie-Lines=	15		param.3	-3,682		param.8	0,31		param.11/14/17:	1275,33	172,9871	0	http://hdl.handle.net/10045/51725				
Model used:	3	1: NRTL	param.4	6,053		param.9	0,30		param.12/15/18:	1549,14	-617,2687	0	Test 3: LLE TERNARY SYSTEM (ALTERNATIVE MODEL)				
		2: UNIQUAC	param.5	-0,693					param.19	0			NRTL with T dependence (15 parameters)				
		3: OTHER	param.6	2,732									$\tau_{ij} = a_{ij} + b_{ij}/T$				
Experimental Data													Calculated Data				
fase 1			fase 2			fase 1			fase 2								
x1	x2	x3	x1	x2	x3	x1	x2	x3	x1	x2	x3						
0,886	0,000	0,114	0,031	0,000	0,969							a _{ij} b _{ij}					
0,814	0,056	0,130	0,047	0,101	0,852							i,j=1,2 0,281 -23,551					
0,758	0,099	0,143	0,064	0,177	0,759							i,j=2,1 -2,710 1275,33					
0,707	0,139	0,154	0,085	0,242	0,674							i,j=1,3 -3,682 1549,14					
0,659	0,177	0,164	0,108	0,296	0,597							i,j=3,1 6,053 -938,986					
0,614	0,212	0,174	0,134	0,339	0,527							i,j=2,3 -0,693 172,9871					
0,569	0,247	0,184	0,163	0,372	0,466							i,j=3,2 2,732 -617,2687					
0,524	0,280	0,196	0,194	0,394	0,412							ALFA12= 0,20					
0,476	0,314	0,210	0,226	0,408	0,366							ALFA13= 0,31					
0,430	0,345	0,225	0,261	0,412	0,327							ALFA23= 0,30					
0,405	0,363	0,232	0,330	0,392	0,270							Plait point (visual location)					
0,390	0,377	0,244	0,374	0,379	0,247							x1 x2 x3					
0,361	0,386	0,254	0,378	0,378	0,243							0,3056 0,4198 0,2746					
0,389	0,374	0,237	0,370	0,383	0,247							I&ECR. 2012, 51(13), 5098-5102.					
0,395	0,373	0,233	0,379	0,380	0,241							http://dx.doi.org/10.1021/ie202793r					

Just to remark the notation used for Aspen Plus in the case of the UNIQUAC model: $\tau_{ji} = \exp(a_{ji} + b_{ji}/T + \dots) = \exp(-u_{ji}/RT) = \exp(-(g_{ji}-g_{ii})/RT) = \exp(-A_{ji}/RT)$.

➤ **exampleLL_test0T.xls**: LLE ternary system. NRTL model. Without exp. and calculated data (to analyze only the corresponding parameters and model).

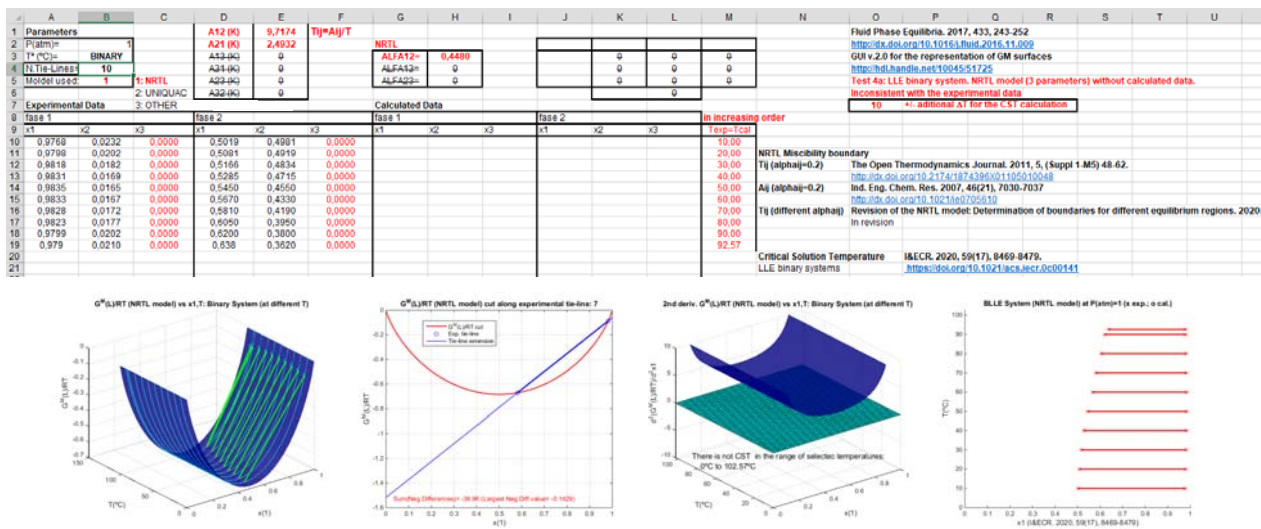
A	B	C	D	E	F	G	H	I	J	K	L	M
Parameters												
P(atm)=	1		T12	-0,27195496	Tij=Aij/T	NRTL						
T* (°C)=	25		T21	0,29349957		ALFA12=	0,2					
N.Tie-Lines=	0		T13	0,42397982		ALFA13=	0,2					
Model used:	1	1: NRTL	T31	0,2811906		ALFA23=	0,2					
		2: UNIQUAC	T23	5,46436472								
		3: OTHER	T32	1,62035756								
Experimental Data												
fase 1			fase 2			fase 1			fase 2			
x1	x2	x3	x1	x2	x3	x1	x2	x3	x1	x2	x3	

LLE BINARY SYSTEMS

As commented previously, the proposed GMcal_TieLinesLL GUI can also be used to represent additionally the evolution of a **binary system** at different temperatures (max. 40 points), including the location of the **critical solution temperature** (UCST, LCST or CML) if there exists (Ref.: Ind. Eng. Chem. Res. 2020, 59, 17, 8469–8479. <https://doi.org/10.1021/acs.iecr.0c00141>).

In this case, it is necessary to write in cell B3 of the excel file the word **BINARY**, and also an additional column including the temperature of each binary data (in increasing order) has also to be included with location = (M10:M50). Additionally, just in the case of the NRTL or UNIQUAC models, the parameters that have to be introduced are the A_{ji} (K) parameters (temperature independent variables). In cell O6 of the corresponding excel file, it is possible to include an additional ΔT for the Critical Solubility Temperature calculation in the range of the experimental temperatures (by defect this $\Delta T=10$).

➤ **exampleLL_test4a.xls**: NRTL model for the binary system (3 parameters) without calculated data.

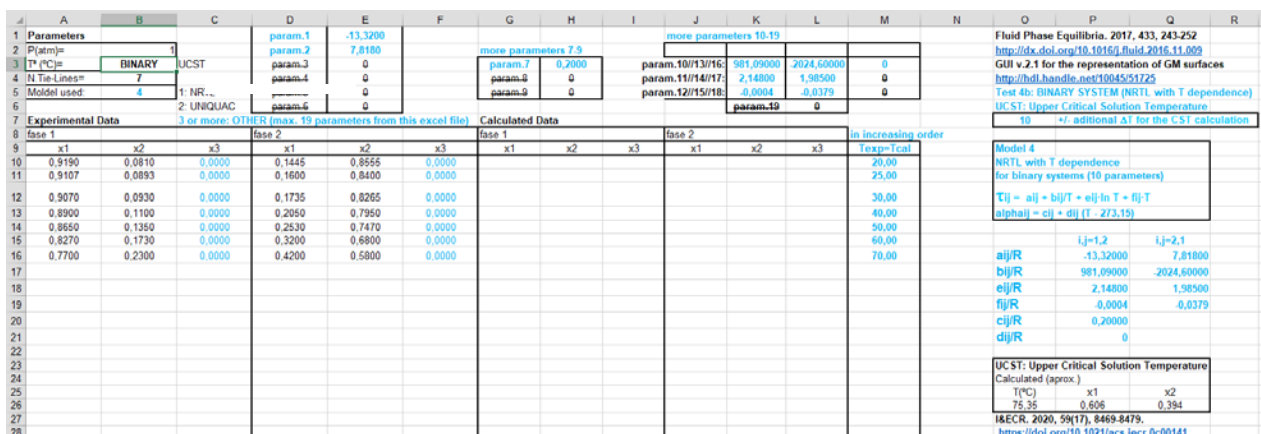


➤ **exampleLL_test4b-d.xls**: Using as “other model (max. 19 parameters directly from the excel file)”, the NRTL model for a binary system with temperature dependent parameters (10 parameters):

$$\tau_{ji} = a_{ji} + b_{ji}/T + e_{ji} \cdot \ln T + f_{ji} \cdot T$$

$$\alpha_{ji} = c_{ji} + d_{ji} (T - 273.15)$$

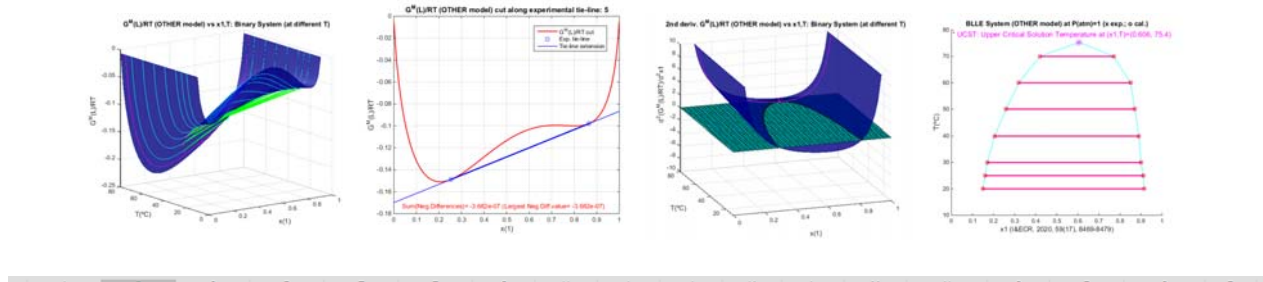
- This “other model” has been numbered in the ModelGExcess.m file as **Model 4**.



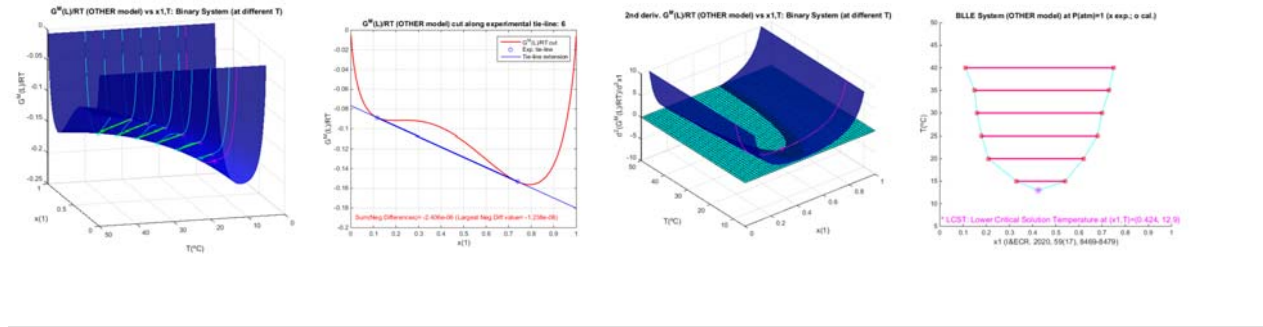
GMcal_TieLinesLL: GUI for the Topological Analysis of Calculated $G^{M(L)}$ Surfaces and Curves, including Tie-Lines, Hessian Matrix, Spinodal Curve, Critical Point Location, etc. for Binary and Ternary LLE Data

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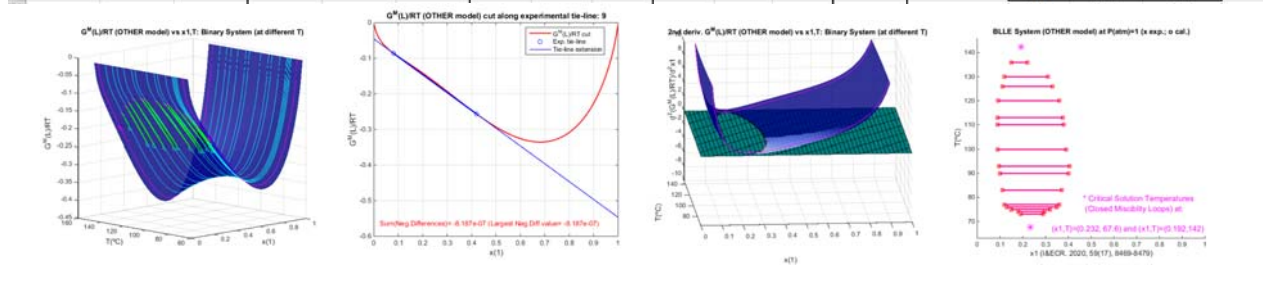
<http://hdl.handle.net/10045/51725>



Parameters		param.1 -55.6300			more parameters 7.9			more parameters 10.19			Fluid Phase Equilibria. 2017, 433, 243-252	
1	P(اتم)=	1	param.2	-102.7000	more parameters 7.9			more parameters 10.19			http://dx.doi.org/10.1016/j.fluid.2016.11.009	
2	T* (°C)=	BINARY	param.3	0	param.7	0.2000	param.10/13/16:	6.30280	6.54080	0	GUI v.2.1 for the representation of GM surfaces	
3	N Tie-Lines=	6	param.4	0	param.8	0	param.11/14/17:	20.49000	21.04000	0	http://hdl.handle.net/10045/51725	
4	M(مدل) used=	4	param.5	0	param.9	0	param.12/15/18:	0.0685	0.0511	0	Test 4c: BINARY SYSTEM (NRTL with T dependence)	
5			param.6	0							LCST: Lower Critical Solution Temperature	
6											LCST: Lower Critical Solution Temperature	
7	Experimental Data	3 or more: OTHER (max. 19 parameters from this excel file)			Calculated Data			param.19			10 +/- additional ΔT for the CST calculation	
8	Base 1	Base 2			Base 1			Base 2			Model 4	
9	x1	x2	x3	x1	x2	x3	x1	x2	x3	Texp=Tcal	NRTL with T dependence	
10	0.5300	0.4700	0.0000	0.3050	0.6950	0.0000				15.00	for binary systems (10 parameters)	
11	0.6250	0.3750	0.0000	0.2250	0.7750	0.0000				20.00	T _{ij} = a _{ij} + b _{ij} T + e _{ij} ln T + f _{ij} T	
12	0.6800	0.3200	0.0000	0.1820	0.8180	0.0000				25.00	alpha _{ij} = c _{ij} + d _{ij} (T - 273.15)	
13	0.7000	0.3000	0.0000	0.1550	0.8450	0.0000				30.00		
14	0.7300	0.2700	0.0000	0.1330	0.8670	0.0000				35.00		
15	0.7400	0.2600	0.0000	0.1170	0.8830	0.0000				40.00		
16												
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Parameters		param.1 -122.1000			more parameters 7.9			more parameters 10.19			Fluid Phase Equilibria. 2017, 433, 243-252	
1	P(اتم)=	1	param.2	-147.3000	more parameters 7.9			more parameters 10.19			http://dx.doi.org/10.1016/j.fluid.2016.11.009	
2	T* (°C)=	BINARY	param.3	0	param.7	0.2000	param.10/13/16:	-5.33160	-5.56700	0	GUI v.2.1 for the representation of GM surfaces	
3	N Tie-Lines=	15	param.4	0	param.8	0	param.11/14/17:	25.04000	30.01000	0	http://hdl.handle.net/10045/51725	
4	M(مدل) used=	4	param.5	0	param.9	0	param.12/15/18:	0.0722	0.0712	0	Test 4d: BINARY SYSTEM (NRTL with T dependence)	
5			param.6	0							Closed Miscibility Loops	
6											10 +/- additional ΔT for the CST calculation	
7	Experimental Data	3 or more: OTHER (max. 19 parameters from this excel file)			Calculated Data			In increasing order			Model 4	
8	Base 1	Base 2			Base 1			Base 2			NRTL with T dependence	
9	x1	x2	x3	x1	x2	x3	x1	x2	x3	Texp=Tcal	for binary systems (10 parameters)	
10	0.2900	0.7100	0.0000	0.1900	0.8100	0.0000				73.00	T _{ij} = a _{ij} + b _{ij} T + e _{ij} ln T + f _{ij} T	
11	0.2950	0.7050	0.0000	0.1800	0.8200	0.0000				74.00	alpha _{ij} = c _{ij} + d _{ij} (T - 273.15)	
12	0.3200	0.6800	0.0000	0.1500	0.8500	0.0000				75.00		
13	0.3500	0.6500	0.0000	0.1300	0.8700	0.0000				76.00		
14	0.3600	0.6400	0.0000	0.1200	0.8800	0.0000				77.00		
15	0.3700	0.6300	0.0000	0.1100	0.8900	0.0000				83.00		
16	0.4200	0.5800	0.0000	0.1000	0.9000	0.0000				90.00		
17	0.4250	0.5750	0.0000	0.0950	0.9050	0.0000				93.00		
18	0.4200	0.5800	0.0000	0.0800	0.9200	0.0000				100.00		
19	0.3800	0.6200	0.0000	0.0890	0.9110	0.0000				110.00		
20	0.3750	0.6250	0.0000	0.0895	0.9105	0.0000				113.00		
21	0.3600	0.6400	0.0000	0.0920	0.9080	0.0000				120.00		
22	0.3300	0.6700	0.0000	0.1100	0.8900	0.0000				126.00		
23	0.3100	0.6900	0.0000	0.1200	0.8800	0.0000				130.00		
24	0.2200	0.7800	0.0000	0.1500	0.8500	0.0000				136.00		
25												
26												
27												
28												
29												
30												
31												
32												
33												



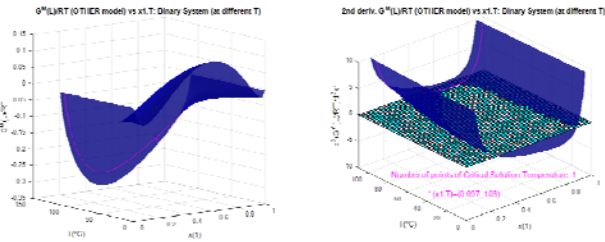
➤ **exampleLL_test5.xls:** Using as “other model (max. 19 parameters directly from the excel file)”, the UNIQUAC model for a binary system with temperature dependent parameters (13 parameters):

$$\tau_{ji} = \exp(a_{ji} + b_{ji}/T + c_{ji} \cdot \ln T + d_{ji} \cdot T)$$

Just to remark the notation used for Aspen Plus in the case of the UNIQUAC model: $\tau_{ji} = \exp(a_{ji} + b_{ji}/T + \dots) = \exp(-u_{ji}/RT) = \exp(-(g_{ji}-g_{ii})/RT) = \exp(-A_{ji}/RT)$.

This “other model” has been numbered in the ModelGExcess.m file as Model 5.

Parameters			Experimental Data			Calculated Data			in increasing order																																								
P(atm)=	1		1			1			1																																								
T* (°C)=	BINARY	UCST																																															
N.Tie-Lines=	0	1: NRTL																																															
Model used:	5	2: UNIQUAC																																															
<table border="1"> <tr> <td>param.1</td> <td>13,30</td> </tr> <tr> <td>param.2</td> <td>-172,62</td> </tr> <tr> <td>param.3</td> <td>-2423,78</td> </tr> <tr> <td>param.4</td> <td>8782,76</td> </tr> <tr> <td>param.5</td> <td>0,20</td> </tr> <tr> <td>param.6</td> <td>23,88</td> </tr> </table>			param.1	13,30	param.2	-172,62	param.3	-2423,78	param.4	8782,76	param.5	0,20	param.6	23,88	<table border="1"> <tr> <td>more parameters 7:9</td> <td></td> </tr> <tr> <td>param.7</td> <td>-0,023</td> </tr> <tr> <td>param.8</td> <td>0,0197</td> </tr> <tr> <td>param.9</td> <td>0</td> </tr> </table>			more parameters 7:9		param.7	-0,023	param.8	0,0197	param.9	0	<table border="1"> <tr> <td>more parameters 10:19</td> <td></td> </tr> <tr> <td>ri</td> <td>qi</td> <td>q1</td> </tr> <tr> <td>param.10/13/16:</td> <td>1,5</td> <td>1,6</td> <td>1,6</td> </tr> <tr> <td>param.11/14/17:</td> <td>3,2</td> <td>2,3</td> <td>2,3</td> </tr> <tr> <td>param.12/15/18:</td> <td>0</td> <td>0</td> <td>0</td> </tr> <tr> <td>param.19</td> <td>10</td> <td>=2</td> <td></td> </tr> </table>			more parameters 10:19		ri	qi	q1	param.10/13/16:	1,5	1,6	1,6	param.11/14/17:	3,2	2,3	2,3	param.12/15/18:	0	0	0	param.19	10	=2	
param.1	13,30																																																
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param.11/14/17:	3,2	2,3	2,3																																														
param.12/15/18:	0	0	0																																														
param.19	10	=2																																															
<p>Remark: qi and q1 are different only for water and some small alcohols</p>																																																	
<p>3 or higher: OTHER (max. 19 parameters from this excel file)</p>			<p>Calculated Data</p>			<p>Remark: qi and q1 are different only for water and some small alcohols</p>			<p>in increasing order</p>																																								
phase 1	x1	x2	x3	phase 2	x1	x2	x3	phase 1	x1	x2	x3	phase 2	x1	x2	x3	Texp=Tcal																																	
																10,00																																	
																100,00																																	



➤ **exampleLL_test0B.xls:** LLE binary system. NRTL model with temperature dependence. Without exp. and calculated data (to analyze only the corresponding parameters and model).

Parameters		Experimental Data		Calculated Data		in increasing order																							
P(atm)=	1																												
T* (°C)=	BINARY																												
N.Tie-Lines=	0																												
Model used:	1	1: NRTL																											
<table border="1"> <tr> <td>A12 (K)</td> <td>1629,200</td> </tr> <tr> <td>A21 (K)</td> <td>483,110</td> </tr> <tr> <td>A13 (K)</td> <td>0</td> </tr> <tr> <td>A31 (K)</td> <td>0</td> </tr> <tr> <td>A23 (K)</td> <td>0</td> </tr> <tr> <td>A32 (K)</td> <td>0</td> </tr> </table>		A12 (K)	1629,200	A21 (K)	483,110	A13 (K)	0	A31 (K)	0	A23 (K)	0	A32 (K)	0	<table border="1"> <tr> <td>Tij=Aij/T</td> <td></td> </tr> </table>		Tij=Aij/T		<table border="1"> <tr> <td>NRTL</td> <td></td> </tr> <tr> <td>ALFA12=</td> <td>0,2000</td> </tr> <tr> <td>ALFA13=</td> <td>0</td> </tr> <tr> <td>ALFA23=</td> <td>0</td> </tr> </table>		NRTL		ALFA12=	0,2000	ALFA13=	0	ALFA23=	0		
A12 (K)	1629,200																												
A21 (K)	483,110																												
A13 (K)	0																												
A31 (K)	0																												
A23 (K)	0																												
A32 (K)	0																												
Tij=Aij/T																													
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ALFA12=	0,2000																												
ALFA13=	0																												
ALFA23=	0																												
<p>3: OTHER</p>		<p>Calculated Data</p>		<p>in increasing order</p>																									
phase 1	x1	x2	x3	phase 2	x1	x2	x3	Texp=Tcal																					
								10,00																					
								100,00																					

**After using this GUI GMcal_TieLinesLL, please complete the following quick inquiry to know your opinion about the graphical user interface developed:



English version
General user



Spanish version
Specific for Spanish students



Facebook

Additional bibliography:

- [1] Olaya, M.M.; Ibarra, I.; Labarta, J.A.; Serrano, M.D.; Marcilla, A. Computing Liquid-Liquid Phase Equilibria: An exercise to understand the nature of false solutions and how to avoid them. *Chemical Engineering Education* **2007**, 41(3), 218-224. Institutional Repository RUA: <http://hdl.handle.net/10045/14277>.
- [2] Marcilla, A.; Labarta J.A.; Serrano, M.D.; Olaya, M.M. Pitfalls on computing liquid-liquid phase equilibria using the k-value method. 11th Mediterranean Congress of Chemical Engineering. EXPOQUIMIA **2008**. Institutional Repository RUA: <http://hdl.handle.net/10045/26610>.
- [3] Marcilla, A.; Labarta, J.A.; Olaya, M.M.; Serrano, M.D. Simultaneous correlation of liquid-liquid, liquid-solid, and liquid-liquid-solid equilibrium data for water + organic solvent + salt ternary systems: hydrated solid phase formation. *Ind. Eng. Chem. Res.* **2008**, 47, 2100-2108. <http://dx.doi.org/10.1021/ie071290w>.
- [4] Reyes, J.A.; Conesa, J.A.; Marcilla, A.; Olaya, M.M. Solid-Liquid Equilibrium Thermodynamics: checking stability in multiphase systems using Gibbs Energy Function. *Ind. Eng. Chem. Res.* **2001**; 40: 902-907. <http://dx.doi.org/10.1021/ie000435v>.
- [5] Olaya, M.M.; Marcilla, A.; Serrano, M.D.; Botella, A.; Labarta, J.A. Simultaneous Correlation of LL, LS and LLS Equilibrium Data for Water + Organic Solvent + Salt Ternary Systems. Anhydrous Solid Phase. *Ind. Eng. Chem. Res.* **2007**, 46(21), 7030-7037. <http://dx.doi.org/10.1021/ie0705610>.
- [6] Marcilla, A.; Olaya, M.M.; Serrano, M.D.; Labarta, J.A. Methods for Improving Models for Condensed Phase Equilibrium Calculations. *Fluid Phase Equilib.* **2010**, 296(1), 15-24. <http://dx.doi.org/10.1016/j.fluid.2009.12.026> (<http://hdl.handle.net/10045/13314>).
- [7] Labarta, J.A.; Olaya, M.M.; Velasco, R.; Serrano, M.D.; Marcilla, A. Correlation of the liquid-liquid equilibrium data for specific ternary systems with one or two partially miscible binary subsystems. *Fluid Phase Equilib.* **2009**, 281, 9-14. <http://dx.doi.org/10.1016/j.fluid.2008.12.002> (<http://hdl.handle.net/10045/24683>).
- [8] Marcilla, A.; Olaya, M.M.; Serrano, M.D.; Velasco, R.; Labarta, J.A. Gibbs energy based procedure for the correlation of type 3 systems including a three-liquid phase region. *Fluid Phase Equilib.* **2008**, 278, 87-95. <http://dx.doi.org/10.1016/j.fluid.2009.04.005> (<http://hdl.handle.net/10045/13315>).
- [9] Olaya, M.M.; Labarta, J.A.; Velasco, R.; Ibarra, I.; Marcilla, A. Modelling Liquid-Liquid Equilibria for Island type Ternary Systems. *Fluid Phase Equilib.* **2008**, 265, 184-191. <http://dx.doi.org/10.1016/j.fluid.2007.12.010> (<http://hdl.handle.net/10045/24681>).
- [10] Marcilla, A.; Olaya, M.M.; Serrano, M.D.; Labarta, J.A. Aspects to be considered for the development of a correlation algorithm for condensed phase equilibrium data of ternary systems. *Ind. Eng. Chem. Res.* **2010**, 49 (20), 10100-10110. <http://dx.doi.org/10.1021/ie1010383>.
- [11] Marcilla, A.; Serrano, M.D.; Labarta, J.A.; Olaya, M.M. Checking Liquid-Liquid Critical Plait Conditions and their Application in Ternary Systems. *Industrial & Engineering Chemistry Research.* **2012**, 51(13), 5098-5102. <http://dx.doi.org/10.1021/ie202793r>.
- [12] Marcilla, A.; Labarta, J.A.; Serrano M.D.; Olaya, M.M. GE Models and Algorithms for Condensed Phase Equilibrium Data Regression in Ternary Systems: Limitations and Proposals. *The Open Thermodynamics Journal.* **2011**, 5, (Suppl 1-M5) 48-62. <http://dx.doi.org/10.2174/1874396X01105010048>.

[13] Olaya, M.M.; Labarta, J.A.; Serrano, M.D.; Marcilla, A. Vapor-Liquid Equilibria using the Gibbs Energy and the Common Tangent Plane Criterion. *Chemical Engineering Education*. **2010**, 44(3), 236-244. Institutional Repository RUA: <http://hdl.handle.net/10045/24677>.

[14] Marcilla, A.; Labarta, J.A.; Olaya, M.M. Gibbs energy of mixing function: topological analysis in azeotropic systems. 27th European Symposium on Applied Thermodynamics ESAT **2014**. Institutional Repository RUA: <http://hdl.handle.net/10045/42249>.

[15] Labarta, J.A.; Olaya, M.M.; Marcilla, A. Mapping Binary Liquid-Vapor or Liquid-Liquid-Vapor Equilibria Regions, including the Different Azeotropic Behaviours, as a Function of the NRTL Binary Parameters. 13th Mediterranean Congress of Chemical Engineering. **2014**. Institutional Repository RUA: <http://hdl.handle.net/10045/42248>.

[16] Marcilla, A.; Labarta, J.A.; Olaya, M.M. Should we trust all the published LLE correlation parameters in phase equilibria? Necessity of their Assessment Prior to Publication. *Fluid Phase Equilibria*. **2017**, 433, 243-252 (<http://dx.doi.org/10.1016/j.fluid.2016.11.009>). RUA: <http://hdl.handle.net/10045/66521>.

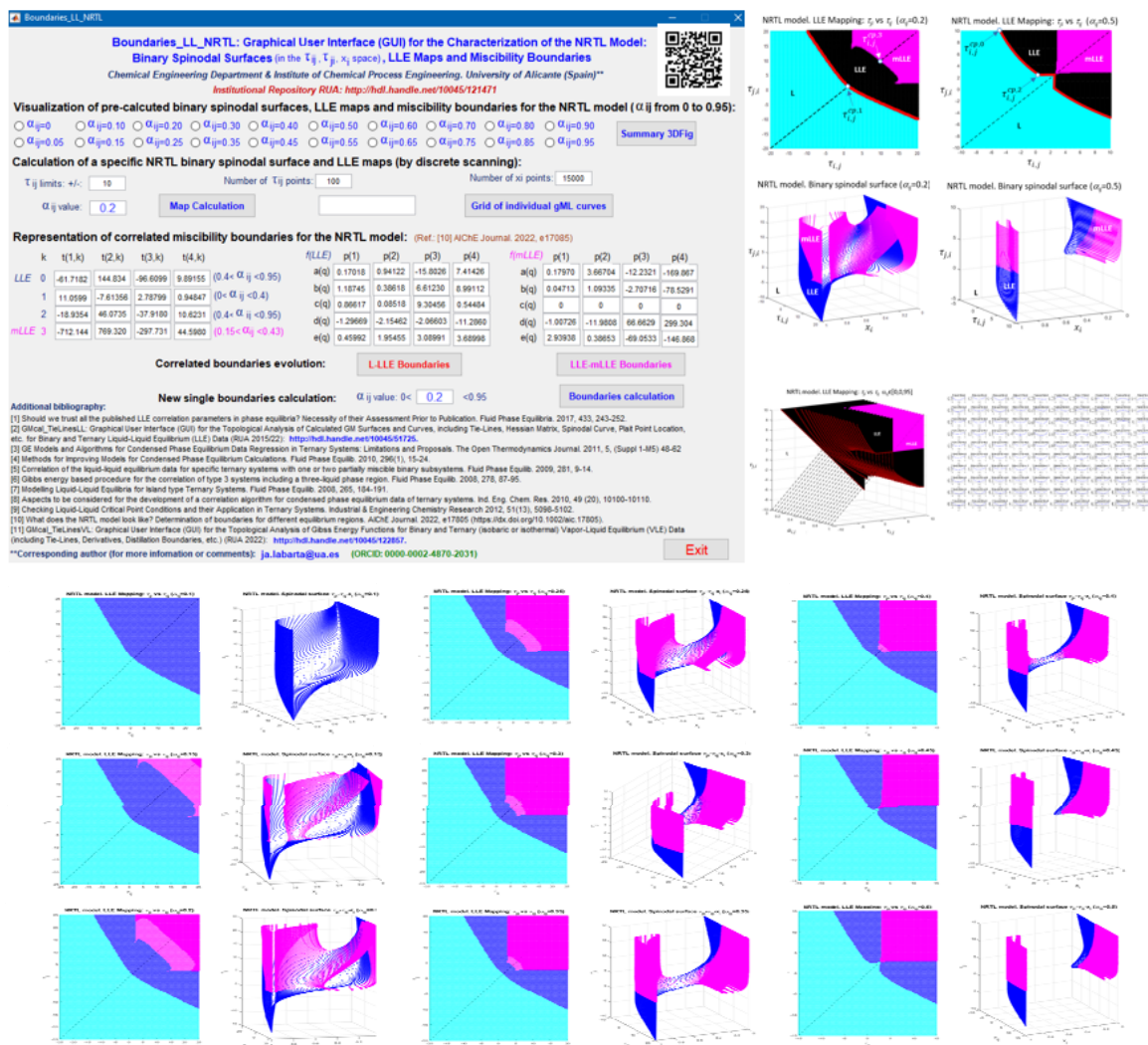
[17] Marcilla, A.; Labarta, J.A.; Olaya, M.M. Phase Equilibrium Correlation. Regarding the full consistency of the thermodynamics. 10th World Congress of Chemical Engineering. Barcelona (SPAIN), 1st-5th October, **2017**. RUA: <http://hdl.handle.net/10045/70035>.

[18] Olaya, M.M.; Carbonell-Hermida, P.; Trives, M.; Labarta, J.A.; Marcilla, A. LLE data correlation using NRTL model for different types of binary systems: UCST, LCST and closed miscibility loops. *Industrial & Engineering Chemistry Research*. 2020, 59(17), 8469-8479. <https://doi.org/10.1021/acs.iecr.0c00141>.

[19] Marcilla, A.; Olaya, M.M.; Labarta, J.A. Ensuring that Correlation Parameters for Liquid-Liquid Equilibrium Produce the Right Results (Guest Editorial by editor-in-chief Joan Brennecke). *Journal of Chemical & Engineering Data*. **2018**, 63(5), 1133-11114 (<https://dx.doi.org/10.1021/acs.jced.8b00260>).

[20] Labarta, J.A.; Olaya, M.M.; Marcilla, A. What does the NRTL model look like? Determination of boundaries for different equilibrium regions. *AIChE Journal*. **2022**, e17805 (<https://dx.doi.org/10.1002/aic.17805>).

[21] Labarta, J.A.; Olaya, M.M.; Marcilla, A. **Boundaries_LL_NRTL**: Graphical User Interface (GUI) for the Characterization of the NRTL Model: Binary Spinodal Surfaces (in the $\tau_{i,j}-\tau_{j,i}-x_i$ space), LLE maps and Miscibility Boundaries. Institutional Repository of the University of Alicante (RUA). **2022**. Publicly available online at: <http://hdl.handle.net/10045/121471>.



[22] Vicente-Martínez, M.; Labarta, J.A. Análisis topológico del modelo NRTL en sistemas binarios y ternarios para caracterizar la tipología de los sistemas líquido-líquido que predice en función de los valores de los parámetros de interacción binaria. Institutional Repository of the University of Alicante (RUA) 2021. Available at: <http://hdl.handle.net/10045/117665>.

[23] Labarta, J.A.; Olaya, M.M.; Marcilla, A. **GMcal_TieLinesVL: Graphical User Interface (GUI) for the Topological Analysis of Experimental and Calculated G^M Functions for Binary and Ternary (isobaric or isothermal) Vapor-Liquid Equilibrium (VLE) data (including Tie-Lines, Derivatives, Distillation Boundaries, LL Critical Points Location, etc.).** Institutional Repository of the University of Alicante (RUA). 2022. Publicly available online at: <http://hdl.handle.net/10045/122857>.

GMcal_TieLinesVL (v.2.2): TOPOLOGICAL ANALYSIS OF VAPOR-LIQUID EQUILIBRIUM DATA AND CORRELATIONS
 Graphical User Interface (GUI) for the Representation of Experimental and Calculated GM Functions for Binary and Ternary (isobaric or isothermal) Vapor-Liquid Equilibrium (VLE or VLLE) data (including Tie-Lines, Derivatives, Distillation Boundaries, LL Critical Points, etc.)
 --- using NRTL, UNIQUAC or an alternative model ---
 Chemical Engineering Department & Institute of Chemical Process Engineering, University of Alicante (Spain)**
 Institutional Repository RUA: <http://hdl.handle.net/10045/122857>

Data source: Introduce Excel file name and press enter: Data Loaded!!

Remark, parameters: NRTL: Aij (K) and $\alpha_{ij} = \alpha_j$; UNIQUAC: Aij (K); n; q; qi and Z

Graphical and topological analysis: Ref: [8,9]

Comparatives of GM/RT and d(GM/RT)/dx1 calculated with models and with the experimental data

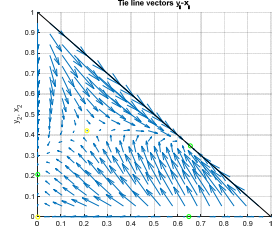
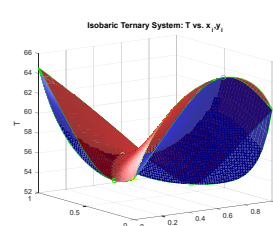
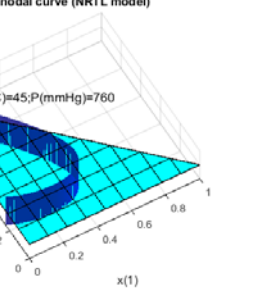
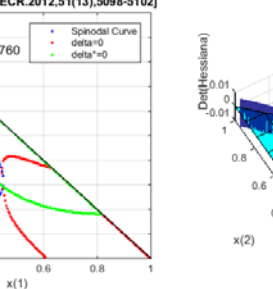
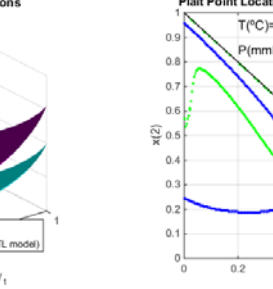
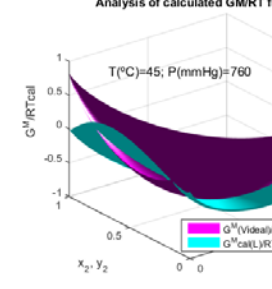
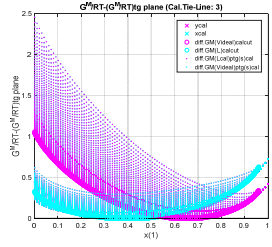
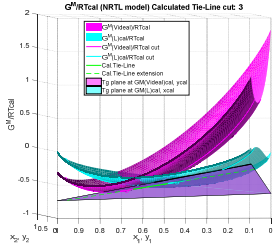
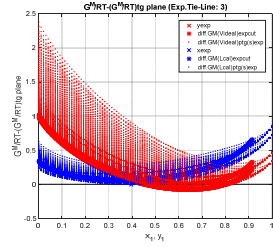
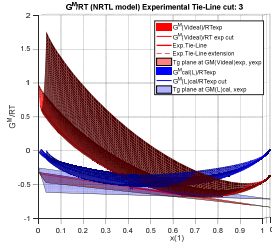
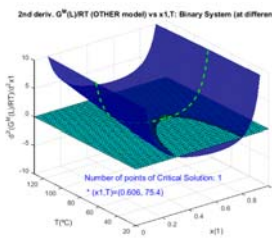
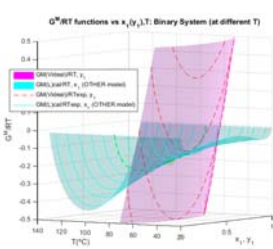
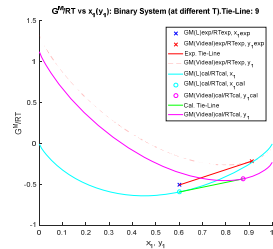
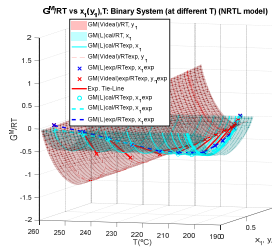
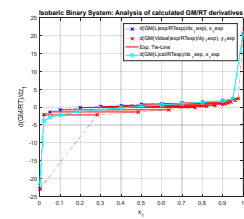
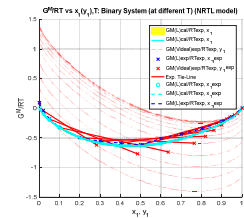
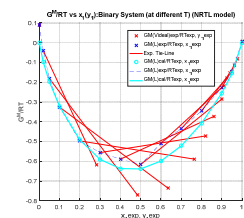
Ref: [10, 11]

NRTL Binary Parameter Boundaries (τ_{ij} vs τ_{ij}) for Total and Partial Miscibility:
 Ref: [5] The Open Thermodynamics Journal, 2011, 5, 48-62. <http://hdl.handle.net/10045/19865>
Ref: [6] AIChE Journal, 2022, e17085. [7] GUI Boundaries_LL_NRTL: <http://hdl.handle.net/10045/121471>

GM/RT functions along selected tie-line: Number of the Tie-line to represent: (press enter)

Additional bibliography:
 [1] Should we trust all the published LLE correlation parameters in phase equilibria? Necessity of their Assessment Prior to Publication. Fluid Phase Equilibria. 2017, 433, 243-252.
 [2] GMcal_TieLinesLL: Graphical User Interface (GUI) for the Topological Analysis of Calculated GM Surfaces and Curves, including Tie-Lines, Hessian Matrix, Spinodal Curve, Plat Point Location, etc. for Binary and Ternary Liquid-Liquid Equilibrium (LLE) Data (RUA 2015/22): <http://hdl.handle.net/10045/51725>
 [3] The unavoidable necessity of considering temperature dependence of the liquid Gibbs energy of mixing for certain VLE data correlations Fluid Phase Equilibria 2016, 473, 17-31.
 [4] Procedure for the correlation of normal appearance VLE data, where the classical models dramatically fail with no apparent reason. Fluid Phase Equilibria. 2019, 423, 88-101.
 [5] GE Models and Algorithms for Condensed Phase Equilibrium Data Regression in Ternary Systems: Limitations and Proposals. The Open Thermodynamics Journal. 2011, 5, 48-62.
 [6] What does the NRTL model look like? Determination of boundaries for different fluid phase equilibrium regions. AIChE Journal, 2022, e17805 (<https://dx.doi.org/10.1002/aic.17805>).
 [7] Boundaries_LL_NRTL: Graphical User Interface (GUI) for the Characterization of the NRTL Model Binary Spinodal Surfaces (in the tau_{ij}-tau_{ij}-xi space), LLE maps and Miscibility Boundaries. (RUA 2022): <http://hdl.handle.net/10045/121471>
 [8] Checking Liquid-Liquid Critical Point Conditions and their Application in Ternary Systems. Industrial & Engineering Chemistry Research. 2012, 51(13), 5098-5102.
 [9] LLE data correlation using NRTL model for different types of binary systems: UCST, LCST and closed miscibility loops. Industrial & Engineering Chemistry Research. 2020, 59(17), 8469-8479.
 [10] Approximate Calculation of Distillation Boundaries for Ternary Azeotropic Systems. Ind. Eng. Chem. Res. 2011, 50 (12), 7462-7466.
 [11] Numerical Determination of Distillation Boundaries for Multicomponent Homogeneous and Heterogeneous Azeotropic Systems. Computer Aided Chemical Engineering. 2010, 28(C), 643-648.

**Corresponding author (for more information or comments): ja.labarta@ua.es (ORCID: 0000-0002-4870-2031)



[24] Labarta, J.A.; Caballero, J.A. **ParamIni_LL_NRTL**: Graphical User Interface (GUI) for the Selection of NRTL Initial Parameters for the Correlation of Ternary Liquid-Liquid Equilibrium Data (Type I, II, III and 0 (LL island), i.e. with 1, 2, 3 or 0 binary pairs partially miscible). Institutional Repository of the University of Alicante (RUA). 2022. Available at: <http://hdl.handle.net/10045/130017>.

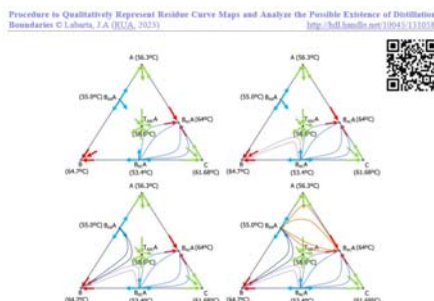
The screenshot shows the ParamIni_LL_NRTL GUI. It has a title bar and a main window containing a QR code and a data source input field. Below this is a table for NRTL parameters with columns for l_j , T_j , and T_j . The table contains three rows of data. There are several buttons for calculations and analysis. At the bottom, there is a list of references and a footer with the author's name and ORCID.

[25] Labarta, J.A.; Caballero, J.A.; Marcilla, A. Liquid-Liquid Equilibrium Data Correlation: Predicting a robust and consistent set of initial NRTL parameters. Computer Aided Chemical Engineering (ESCAPE 33). 2023, 52, 3457-3462. DOI: <https://doi.org/10.1016/B978-0-443-15274-0.50552-7>. Inst. Rep. of the Univ. of Alicante (RUA) 2023: <http://hdl.handle.net/10045/134753>.

[26] Labarta, J.A. Review of Graphical User Interfaces (GUIs) related to the Correlation of Phase Equilibrium Data. 2023. DOI: <https://doi.org/10.5281/zenodo.7628814>.

[27] Marcilla, A.; Gómez, A., García, A.N.; Beltrán, M.I.; Olaya, M.M.; Labarta, J.A. Operaciones de separación de transferencia de materia. Editorial Síntesis (2022). ISBN: ISBN: 9788413571775.

[28] Labarta, J.A. Procedure to Qualitatively Represent Residue Curve Maps and Analyze the Possible Existence of Distillation Boundaries (RUA, 2023). <http://hdl.handle.net/10045/131058>.



[29] Gómez, A.; Ruiz, F.; Marcilla, A.; Labarta, J.A.; Menargues, S. Diseño de la separación de mezclas ternarias (I). Conceptos gráficos del equilibrio entre fases. Ingeniería Química. 2001, 377, 219-229. Institutional Repository (RUA): <http://hdl.handle.net/10045/24715>.

[30] Gómez, A.; Ruiz, F.; Marcilla, A.; Labarta, J.A.; Menargues, S. Diseño de la separación de mezclas ternarias (II). Aplicación de conceptos gráficos a la separación de mezclas azeotrópicas.

Ingeniería Química. 2001, 379, 253-262. Institutional Repository (RUA): <http://hdl.handle.net/10045/24716>.

[31]I. Díaz, M- Rodríguez, E.J. González, M. González-Miquel, 2019, A simple and reliable procedure to accurately estimate NRTL interaction parameters from liquid-liquid equilibrium data. Chem. Eng. Sci., 193: 370-378.

[32]M. Glass, M. Aigner, J. Viell, A. Jupke, A. Mitsos, 2017, Liquid-liquid equilibrium of 2-methyltetrahydrofuran/water over wide temperature range: Measurements and rigorous regression. Fluid Phase Equilib., 433, 212-225.

[33]D.L. de Klerk. C.E. Schwarz, 2023, Simplified Approach to the Parameterization of the NRTL Model for Partially Miscible Binary Systems: $\tau\tau$ LLE Methodology, Industrial & Engineering Chemistry Research, 62, 4, 2021-2035.

[34]M. Sapkowski, T. Hofman. 2023, Problems and limitations in the calculation of liquid-liquid equilibrium, Fluid Phase Equilibria, 571, 113823.