

### **Boundaries\_LL\_NRTL**

### Graphical User Interface (GUI) for the Characterization of the NRTL Model: Binary Spinodal Surfaces (in the $\tau_{i,j}$ - $\tau_{j,i}$ - $x_i$ space), LLE Maps, and Miscibility Boundaries

Institutional Repository of the University of Alicante (RUA): <u>http://hdl.handle.net/10045/121471</u>

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**Keywords:** Liquid-Liquid Equilibrium, Phase Equilibria Calculation, Gibbs Energy of Mixing, NRTL, Spinodal Curve, Hessian Matrix, Data Correlation, Binary and Ternary Systems, Tie Line, Thermodynamic Models, Equilibrium Maps, Miscibility Boundaries.

## **1. INTRODUCTION**

In a similar way to the previous MatLab Graphical User Interfaces (GUI's) developed to systematically check the consistency of LLE or VLE data correlation results, <u>GMcal\_TieLinesLL</u> [1,2] <u>GMcal\_TieLinesVL</u> [3] respectively, this Graphical User Interface (Boundaries\_LL\_NRTL) has been developed as a friendly tool for the analysis of the NRTL model, and it is directly related with the *AIChE Journal* research paper: "*What does the NRTL model look like? Determination of boundaries for different fluid phase equilibrium regions*" [4]. This GUI allows the direct visualization and calculation of 3D representations (in the  $\tau_{i,j}$ - $\tau_{j,i}$ - $x_i$  space) and 2D projections (in the  $\tau_{i,j}$ - $\tau_{j,i}$  plane) of binary spinodal surfaces, LLE maps, and miscibility boundaries of the NRTL model for different values of the non-randomness parameter ( $\alpha_{i,j}$ ) between 0 and 0.95.

The analysis of all these figures allows researchers involved in the correlation of experimental liquid-liquid equilibrium data, to establish relations between the typology of the system under study (regarding the behavior of all the binary subsystems) and the values of the NRTL binary interaction parameters  $\tau_{i,j}$ - $\tau_{j,i}$  consistent with that typology.

### NRTL model for non-ideal liquid mixtures

 $\succ Gibbs energy of mixing$   $g^{M(L)} = G^{Mixture,L}/RT = G^{Ideal}/RT + G^{Excess}/RT) = x_i \cdot ln (x_i) + G^E/RT$ 

#### > Excess Gibbs energy:

$$\frac{G^{E(NRTL)}}{RT} = \sum_{i=1}^{C} x_i \cdot \frac{\sum_{j=1}^{C} \tau_{j,i} \cdot G_{j,i} \cdot x_j}{\sum_{k=1}^{C} G_{k,i} \cdot x_k}$$

> Activity coefficient:

$$\ln \gamma_i = \frac{G^E}{RT} + \sum_{j=1}^C x_i \cdot \left(\frac{\partial (G^E/RT)}{\partial x_i} - \frac{\partial (G^E/RT)}{\partial x_j}\right)$$

$$\ln \gamma_i^{NRTL} = \frac{\sum_{j=1}^C \tau_{j,i} \cdot G_{j,i} \cdot x_j}{\sum_{k=1}^C G_{k,i} \cdot x_k} + \sum_{j=1}^C \frac{G_{i,j} \cdot x_j}{\sum_{k=1}^C G_{k,j} \cdot x_k} \cdot \left(\tau_{i,j} - \frac{\sum_{m=1}^C \tau_{m,j} \cdot G_{m,j} \cdot x_m}{\sum_{k=1}^C G_{k,j} \cdot x_k}\right)$$

with  $\tau_{j,i} = \frac{A_{j,i}}{RT}$ ;  $G_{j,i} = exp(-\alpha_{j,i} \cdot \tau_{j,i})$ ;  $A_{i,i}=0$  and  $\alpha_{i,j} = \alpha_{j,i}$ 

For a binary system (C=2):

$$\begin{aligned} \frac{G^{E(NRTL)}}{RT} &= x_1 \cdot \frac{\tau_{21} \cdot G_{21} \cdot x_2}{x_1 + G_{21} \cdot x_2} + x_2 \cdot \frac{\tau_{12} \cdot G_{12} \cdot x_1}{G_{12} \cdot x_1 + x_2} \\ \frac{\partial (G^{E(NRTL)}/RT)}{\partial x_1} &= \frac{\tau_{21} \cdot (G_{21} \cdot x_2)^2}{(x_1 + G_{21} \cdot x_2)^2} + \frac{\tau_{12} \cdot G_{12} \cdot x_2^2}{(G_{12} \cdot x_1 + x_2)^2} \\ \frac{\partial (G^{E(NRTL)}/RT)}{\partial x_2} &= \frac{\tau_{21} \cdot G_{21} \cdot x_1^2}{(x_1 + G_{21} \cdot x_2)^2} + \frac{\tau_{12} \cdot (G_{12} \cdot x_1)^2}{(G_{12} \cdot x_1 + x_2)^2} \\ \ln \gamma_1^{NRTL} &= x_2^2 \cdot \left[ \frac{\tau_{21} \cdot G_{21}^2}{(x_1 + x_2 \cdot G_{21})^2} + \frac{\tau_{12} \cdot G_{12}}{(x_2 + x_1 \cdot G_{12})^2} \right] \\ \ln \gamma_2^{NRTL} &= x_1^2 \cdot \left[ \frac{\tau_{12} \cdot G_{21}^2}{(x_2 + x_1 \cdot G_{12})^2} + \frac{\tau_{21} \cdot G_{21}}{(x_1 + x_2 \cdot G_{21})^2} \right] \end{aligned}$$

## 2. USER INSTRUCTIONS

#### 1. Download instructions

- 1. Download the file to your computer in a known folder: **Boundaries\_LL\_NRTL.zip**
- 2. Unzip the file

#### 2. <u>Using the GUI Boundaries\_LL\_NRTL</u>

- 1. Open Matlab software
- 2. Once in MatLab, <u>select</u> the folder where the file **Boundaries\_LL\_NRTL.zip** was unzipped <u>as "current folder"</u>.
- 3. Localize and execute the file Boundaries\_LL\_NRTL.p from the MatLab Command Window (i.e.: writing Boundaries\_LL\_NRTL in the Command Window, and pressing enter). The next windows will appear.

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\ <i>I</i>	Boundaries_LL_NRTL: Graphical User Interface (GUI) for the Characterization of the NRTL Model: Binary Spinodal Surfaces (in the T <sub>ij</sub> , T <sub>jj</sub> , x <sub>i</sub> space), LLE Maps and Miscibility Boundaries Chemical Engineering Department & Institute of Chemical Process Engineering. University of Alicante (Spain)** Institutional Repository RUA: http://hdl.handle.net/10045/121471																	
visualiz																0.95):		
$\bigcirc \alpha_{ij=0} \qquad \bigcirc \alpha_{ij=0,1} \\ \bigcirc \alpha_{ij=0,05} \qquad \bigcirc \alpha_{ij=0,1} \\ \bigcirc \alpha_{$			$\bigcirc \alpha_{ij=0.20} \bigcirc \alpha_{ij=0.30} \bigcirc \alpha_{ij=0.30} \bigcirc \alpha_{ij=0.35} $			$\bigcirc \alpha_{ij=0.40}$	Oui	$\alpha_{ij=0.50} \bigcirc \alpha_{ij=0.60} \bigcirc \alpha_{ij=0.60} \bigcirc \alpha_{ij=0.60} \bigcirc \alpha_{ij=0.65} \bigcirc $			$0^{0}$ $u_{ij=0.70}$ $0^{0}$ $u_{ij=0.80}$ $0^{0}$			0.90	Summary 3DFig			
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T ij limits: +/-: 10 Number of Uj points: 100 Number of Xi points. 15000																		
αį	j value:	0.2		Map Calculation							Grid	Grid of individual gML curves						
Repres	Representation of correlated miscibility boundaries for the NRTL model: (Ref.: [10] AIChE Journal. 2022, e17085)																	
k	t(1,k)	t(2,k)	t(3,k)	t(4,k)	•		f(LLE)	p(1)	p(2)	p(3)	p(4)	f(mLLE	) p(1)	p(2)	p(3)	p(4)		
	61 7182	144 834	96 6099	9.89155	(0.4< α j	i <0.95)	a(q)	0.17018	0.94122	-15.8026	7.41426	a(q)	0.17970	3.66704	-12.2321	-169.867		
1	11 0500	-7.61356	2 78799	0.94847	(0< α ii -	<0.4)	b(q)	1.18745	0.38618	6.61230	8.99112	b(q)	0.04713	1.09335	-2.70716	-78.5291		
2	-18,9354	46.0735	-37.9180	10.6231	(0 4< α i	i <0.95)	c(q)	0.86617	0.08518	9.30456	0.54484	c(q)	0	0	0	0		
mLLE 3	-712.144	769.320	-297.731	44.5980	(0.15<0	ii <0.43)	d(q)	-1.29669	-2.15462	-2.06603	-11.2860	d(q)	-1.00726	-11.9808	66.6629	299.304		
					],	· · ·	e(q)	0.45992	1.95455	3.08991	3.68998	e(q)	2.93938	0.38653	-69.0533	-146.868		
	Correlated boundaries evolution: L-LLE Boundaries LLE-mLLI															Boundaries		
			Ne	wainal	e heund		ulation	. α ii	ivaluo: 0<	0.2	<0.95		Bo	undarios	calculat	tion		
Additional	bibliograp	ohy:	Ne	w singi	e bound	laries calc	ulation	. wij	value. 0<	0.2	<0.55			unuaries	carculat			
1] Should w 21 GMcal Ti	e trust all t eLinesLL: (	he publishe Graphical U	d LLE corre ser Interfac	lation parar e (GUI) for	meters in ph the Topoloc	nase equilibria? pical Analysis o	Necessit of Calculat	y of their A ted GM Sur	ssessment	Prior to Pub urves, inclu	olication. Flui uding Tie-Lin	d Phase Equili es. Hessian N	bria. 2017, 4 Iatrix. Spino	433, 243-25 Idal Curve, I	i2. Plait Point Le	ocation.		
etc. for Bina	ry and Terr	hary Liquid-	Liquid Equili	brium (LLE)	) Data (RUA	2015/22): htt	p://hdl.h	andle.net/	10045/5172	5.	The Ones	Thermodule	ice loureel	2014 5 (5	Suppl ( ME)	49.60		
4] Methods	for Improvi	ng Models f	or Condens	ed Phase Equil	quilibrium C	alculations. Flu	id Phase	Equilib. 201	0, 296(1), 1	5-24.	s. The Open	rnernodynan	ics Journal	. 2011, 5, (3	suppi i-mo)	40-02		
5] Correlation 6] Gibbs en	on of the liq ergy based	uid-liquid ec I procedure	for the cor	ta for spec relation of t	ific ternary ype 3 syste	systems with o ems including a	one or tw three-liqu	o partially r Jid phase re	niscible bina egion. Fluid I	ry subsyst Phase Equil	ems. Fluid Ph ib. 2008, 278	1ase Equilib. 2 3, 87-95.	009, 281, 9	-14.				
[7] Modelling [8] Aspects	Liquid-Liqu to be consi	uid Equilibria	for Island t	ype Ternar ient of a co	y Systems.	Fluid Phase Eq	uilib. 200 densed p	8, 265, 184 hase equili	-191. brium data o	f ternary sy	vstems Ind	Ena Chem Re	es 2010 49	9 (20) 1010	0-10110			
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[11] GMcal_	TieLinesVL	: Graphical	User Interfa	ce (GUI) fo	or the Topol	ogical Analysis	of Gibss	Energy Fu	nctions for I	Binary and	Ternary (isol	baric or isothe	rmal) Vapo	r-Liquid Equ	uilibrium (VL	E) Data		
including Tie	e-Lines, De	rivatives, D	istillation Bo	undaries, e	nc.) (RUA 2	:022): http://h	di.handi	e.net/1004	15/122857.	002 4070	2024)					Ex	dit	

- 4. Once the main window of the GUI appears, it is possible to create different diagrams by using the corresponding push buttons:
  - > Pre-calculated 3D binary spinodal surfaces (in the  $\tau_{i,j}$ - $\tau_{j,i}$ - $x_i$  space) and miscibility boundaries (2D projections in the  $\tau_{i,j}$ - $\tau_{j,i}$  plane) for the NRTL model for different  $\alpha_{i,j}$  values (from 0 to 0.95). (e.g. Figures 1a).
  - Figure 3D resume with all the 2D projections for different  $\alpha_{i,j}$  values, from 0 to 0.95 (Figure 1b).
  - Calculation and representation of a specific NRTL binary spinodal surface and its 2D projection for a concrete value of  $\alpha_{i,j}$  (by discrete scanning of the binary parameters  $\tau_{i,j}$  and  $\tau_{j,i}$ ). It is possible to choose the  $\alpha_{i,j}$  value, the upper/lower limits of the tau parameters, and the number of tau and x points used in the corresponding scanning. The default values are respectively: 0.2; +/- 10; 100 and 15000. (e.g. Figures 1c)

Remark: The possibility of the existence of LL phase splitting has been analyzed through zeros of the second derivative of the Gibbs energy of mixing (with respect to

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the molar fraction), for each set of  $\tau_{i,j}$ - $\tau_{j,i}$  values, in the whole composition range ( $x_i \in [0,1]$ ). That is due because the presence of a LL splitting requires the existence of two zeros in the second derivative of the Gibbs energy of mixing, as can be observed in the following figures: Case a represents a totally homogeneous binary mixture in the whole range of compositions, while cases b-e correspond to binary systems with two partially miscible components just one LLE. When there are more than two compositions with a value of zeros in the second derivative of the Gibbs energy of mixing, more than one LL splitting exists (case f or g):



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- > Calculation and representation of a grid of individual Gibbs energy of mixing curves specific NRTL binary spinodal surface for a concrete value of  $\alpha_{i,j}$  and different pairs of the binary parameters  $\tau_{i,j}$  and  $\tau_{j,i}$ . (e.g. Figure 1d).
- Representation of the evolution of the correlated L-LLE and LLE-mLLE boundaries observed for the NRTL by using the set of parameters included as default corresponding with the following equations and definitions [3]) (Figures 1e).

$$f(\tau_{i,j}, \alpha_{i,j}) \equiv \tau_{j,i}^{cal} = p_1 \cdot \tau_{i,j}^3 + p_2 \cdot \tau_{i,j}^2 + p_3 \cdot \tau_{i,j} + p_4$$

$$p_q = a_q + b_q \cdot ln(\alpha_{i,j} + c_q) + d_q \cdot \alpha_{i,j} + e_q \cdot \alpha_{i,j}^2 \qquad q = \{1,2,3,4\}$$

$$g_k(\alpha_{i,j}) \equiv \tau_{i,j}^{cp,k} = t_{1,k} \cdot \alpha_{i,j}^3 + t_{2,k} \cdot \alpha_{i,j}^2 + t_{3,k} \cdot \alpha_{i,j} + t_{4,k}$$



If we use the value of 10 as the upper bound for  $\tau_{j,i}$ , in both cases, the miscibility boundary (red line) exists in the range of values of  $\tau_{i,j}$  from -5 to 10, i.e.  $\tau_{i,j}c^{p,\theta} = -5$ .

> Calculation and representation of the correlated NRTL miscibility boundaries for a specific value of  $\alpha_{i,j}$  and different pairs of the binary parameters  $\tau_{i,j}$  and  $\tau_{j,i}$ . (e.g. Figure 1f).





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c)

g ML 0.5 .6:T21=1 7:T21=1 12=2.8:T21=1 2=3.9:T21=1 12=5:T21=1. :T21=1 0.5 0.5 0.5 0.5 0.5 0.5 T12=5; T21=0.6 12=2.8:T21=0 9·T21 0.5 0.5 12=5;T21=-0.6 =3.9;T21= 5;T21 3;T21 .7;T21 .6;T21 =0.6;T21= =1.7;T21= =2.8;T21=-2=2.8:T21= :T21: T21=-2 0.5 0.5 0.5 0.5 12=5;T21=-3. 6-T21: 0.6:T21=-3. 2=2.8;T21= =3.9:T21=-3 0.5 0.5 0.5 0.5 0.5 0.5 0.5 ×1 2=-1.7;T21=-×1 =-0.6;T21=-5 ×1 12=0.6;T21=-5 ×1 12=1.7;T21= ×1 12=2.8;T21=-2=3.9;T21=-12=5;T21=gML g ML gML gML 0.5 0.5 0.5 d) LLE-mLLE bound ries: Tij (alphaij=[0.15,0.43] del: LLE-mLLE b L-LLE boundaries: Tij (alphaij=[0,0.95])  $\tau_{ii}$  vs  $\tau_{ii}$  ( $\alpha_{ii}$ =[0.43,0.5 т e) Miscibility boundaries: Tij (alphaij=0.3) 12 10 LE reak Ē Hon L-LLE boundar 2 Tij f)

**Figure 1.** Examples of different individual graphs that can be generated with the GUI developed: Boundaries\_LL\_NRTL (<u>RUA</u>).

# Boundaries\_LL\_NRTL: GUI for the Characterization of the NRTL Model: Binary Spinodal Surfaces (in the τ<sub>i,j</sub>-τ<sub>j,i</sub>-x<sub>i</sub> space), LLE Maps, and Miscibility Boundaries © Labarta, J.A et al. (<u>RUA</u>, 2022) <u>http://hdl.handle.net/10045/121471</u>

A deeper analysis of the mLLE region for  $\alpha_{i,j}$ =0.2, indicates that this region presents two different behaviors (see Figure 2a,b): mLLE(a) where one of the LL splittings is metastable and mLLE(b) where both LL splittings are stable (Sapkowski & Hofman, Fluid Phase Equilibria, 2023). In any case, both regions mLLE(a) and (b) present values of the  $\tau_{i,j}$  too large (larger than 10) in order to correlate correctly LLE data, especially in multicomponent systems, because e.g. in ternary systems they produce Gibbs energy of mixing functions so flattened that produce too much uncertainty in the solutions obtained.

On the other hand, the region mLLE(a) decreases when the  $\alpha_{i,j}$  value increases, and it disappears at an  $\alpha_{ij}$  value around 0.45 (Figure 2c).





**Figure 2.** Evolution of the mLLE region with the  $\alpha_{i,j}$  value.

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#### **Bibliography and additional references:**

[1] Labarta, J.A; Olaya, M.M.; Marcilla, A. **GMcal\_TieLinesLL:** Graphical User Interface (GUI) for Topological Analysis of Calculated G<sup>M</sup> Surfaces and Curves, including Tie-Lines, Hessian Matrix, Spinodal Curve, Plait Point Location, Miscibility Boundaries, etc. for Binary and Ternary Liquid-Liquid Equilibrium (LLE) Data. *Institutional Repository* (RUA) **2015-2022**. Publicly available online at: <u>http://hdl.handle.net/10045/51725</u>. (Including a Thermodynamic Review).



[2]Marcilla, A.; Labarta, J.A.; Olaya, M.M. Should we trust all the published LLE correlation parameters in phase equilibria? Necessity of their Assessment Prior to Publication. Fluid Phase Equilibria. **2017**, 433, 243-252 (<u>http://dx.doi.org/10.1016/j.fluid.2016.11.009</u>). <u>RUA</u>: <u>http://hdl.handle.net/10045/66521</u>. (The 4th most cited article published in Fluid Phase Equilibria in 2017-2020 (extracted from Scopus 11/2020).)

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[3] Labarta, J.A.; Olaya, M.M.; Marcilla, A. **GMcal\_TieLinesVL:** Graphical User Interface (GUI) for the Topological Analysis of Experimental and Calculated G<sup>M</sup> Functions for Binary and Ternary (isobaric or isothermal) Vapor-Liquid Equilibrium (VLE) data (including Tie-Lines, Derivatives, Distillation Boundaries, LL Critical Points Location, etc.). Institutional Repository of the University of Alicante (RUA). **2022**. Available at: http://hdl.handle.net/10045/122857.





[4] Labarta, J.A.; Olaya, M.M.; Marcilla, A. What does the NRTL model look like? Determination of boundaries for different fluid phase equilibrium regions. AIChE Journal. **2022**, e17805 (https://dx.doi.org/10.1002/aic.17805).

[5] Olaya, M.M.; Ibarra, I.; Labarta, J.A.; Serrano, M.D.; Marcilla, A. Computing Liquid-Liquid Phase Equilibria: An exercise to understand the nature of false solutions and how to avoid them. Chemical Engineering Education. **2007**, 41(3), 218-224. Institutional Repository RUA: <u>http://hdl.handle.net/10045/14277</u>.

[6] Marcilla, A.; Labarta J.A.; Serrano, M.D.; Olaya, M.M. Pitfalls on computing liquid-liquid phase equilibria using the k-value method. 11<sup>th</sup> Mediterranean Congress of Chemical Engineering. EXPOQUIMIA **2008**. Institutional Repository RUA: http://hdl.handle.net/10045/26610.

[7] Marcilla, A.; Labarta, J.A.; Olaya, M.M.; Serrano, M.D. Simultaneous correlation of liquidliquid, liquid-solid, and liquid-liquid-solid equilibrium data for water + organic solvent + salt ternary systems: hydrated solid phase formation. *Ind. Eng. Chem. Res.* **2008**, 47, 2100-2108. <u>http://dx.doi.org/10.1021/ie071290w</u>.

[8] Reyes, J.A.; Conesa, J.A.; Marcilla, A.; Olaya, M.M. Solid-Liquid Equilibrium Thermodynamics: checking stability in multiphase systems using Gibbs Energy Function. *Ind. Eng. Chem. Res.* **2001**; 40: 902-907. <u>http://dx.doi.org/10.1021/ie000435v</u>.

[9] Olaya, M.M.; Marcilla, A.; Serrano, M.D.; Botella, A.; Labarta, J.A. Simultaneous Correlation of LL, LS and LLS Equilibrium Data for Water + Organic Solvent + Salt Ternary Systems. Anhydrous Solid Phase. *Ind. Eng. Chem. Res.* **2007**, 46(21), 7030-7037. http://dx.doi.org/10.1021/ie0705610.

[10] Marcilla, A.; Olaya, M.M.; Serrano, M.D.; Labarta, J.A. Methods for Improving Models for Condensed Phase Equilibrium Calculations. *Fluid Phase Equilib.* **2010**, 296(1), 15-24. http://dx.doi.org/10.1016/j.fluid.2009.12.026 (http://hdl.handle.net/10045/13314).

[11] Labarta, J.A.; Olaya, M.M.; Velasco, R.; Serrano, M.D.; Marcilla, A. Correlation of the liquid-liquid equilibrium data for specific ternary systems with one or two partially miscible binary subsystems. *Fluid Phase Equilib.* **2009**, 281, 9-14. http://dx.doi.org/10.1016/j.fluid.2008.12.002 (http://hdl.handle.net/10045/24683).

[12] Marcilla, A.; Olaya, M.M.; Serrano, M.D.; Velasco, R.; Labarta, J.A. Gibbs energy based procedure for the correlation of type 3 systems including a three-liquid phase region. *Fluid Phase Equilib.* **2008**, 278, 87-95. <u>http://dx.doi.org/10.1016/j.fluid.2009.04.005</u> (http://hdl.handle.net/10045/13315).

[13] Olaya, M.M.; Labarta, J.A.; Velasco, R.; Ibarra, I.; Marcilla, A. Modelling Liquid-Liquid Equilibria for Island type Ternary Systems. *Fluid Phase Equilib.* **2008**, 265, 184-191. http://dx.doi.org/10.1016/j.fluid.2007.12.010 (http://hdl.handle.net/10045/24681).

[14] Marcilla, A.; Olaya, M.M.; Serrano, M.D.; Labarta, J.A. Aspects to be considered for the development of a correlation algorithm for condensed phase equilibrium data of ternary systems. *Ind. Eng. Chem. Res*. **2010**, 49 (20), 10100-10110. <u>http://dx.doi.org/10.1021/ie1010383</u>.

[15] Marcilla, A.; Serrano, M.D.; Labarta. J.A.; Olaya, M.M. Checking Liquid-Liquid Critical Plait Conditions and their Application in Ternary Systems. *Industrial & Engineering Chemistry Research.* **2012**, 51(13), 5098-5102. <u>http://dx.doi.org/10.1021/ie202793r</u>.

[16] Marcilla, A.; Labarta, J.A.; Serrano M.D.; Olaya, M.M. GE Models and Algorithms for Condensed Phase Equilibrium Data Regression in Ternary Systems: Limitations and Proposals. *The Open Thermodynamics Journal.* **2011**, 5, (Suppl 1-M5) 48-62. http://dx.doi.org/10.2174/1874396X01105010048.

[17] Olaya, M.M.; Labarta, J.A.; Serrano, M.D.; Marcilla, A. Vapor-Liquid Equilibria using the Gibbs Energy and the Common Tangent Plane Criterion. Chemical Engineering Education. **2010**, 44(3), 236-244. Institutional Repository RUA: <u>http://hdl.handle.net/10045/24677</u>.

[18] Marcilla, A.; Labarta J.A; Olaya, M.M. Gibbs energy of mixing function: topological analysis in azeotropic systems. 27<sup>th</sup> European Symposium on Applied Thermodynamics ESAT **2014**. Institutional Repository RUA: <u>http://hdl.handle.net/10045/42249</u>.

[19] Labarta, J.A.; Olaya, M.M.; Marcilla, A. Mapping Binary Liquid-Vapor or Liquid-Liquid-Vapor Equilibria Regions, including the Different Azeotropic Behaviours, as a Function of the NRTL Binary Parameters. 13th Mediterranean Congress of Chemical Engineering. **2014**. Institutional Repository RUA: <u>http://hdl.handle.net/10045/42248</u>.

[20] Marcilla, A.; Labarta, J.A.; Olaya, M.M. Refreshing the Relevance of Analyzing the Topology of the GM Function in the Phase Equilibrium Correlation Data. Equifase 2015. X Iberoamerican Conference on Phase Equilibria and Fluid Properties For Process Design. Alicante (SPAIN), 28 June - 1 July, **2015**. (Book of Abstracts: <u>http://hdl.handle.net/10045/50110</u>).

[21] Marcilla, A.; Labarta, J.A.; Olaya, M.M. Phase Equilibrium Correlation. Regarding the full consistency of the thermodynamics. 10th World Congress of Chemical Engineering. Barcelona (SPAIN), 1t-5th October, **2017**. RUA: <u>http://hdl.handle.net/10045/70035</u>.

[22] Olaya, M.M.; Carbonell-Hermida, P.; Trives, M.; Labarta, J.A.; Marcilla, A. LLE data correlation using NRTL model for different types of binary systems: UCST, LCST and closed miscibility loops. Industrial & Engineering Chemistry Research. 2020, 59(17), 8469-8479. https://doi.org/10.1021/acs.iecr.0c00141.

[23] Marcilla, A.; Olaya, M.M.; Labarta, J.A. Ensuring that Correlation Parameters for Liquid-Liquid Equilibrium Produce the Right Results (Guest Editorial by editor-in-chief Joan Brennecke). Journal of Chemical & Engineering Data. **2018**, 63(5), 1133-11114 (https://dx.doi.org/10.1021/acs.jced.8b00260).

[24] Rodríguez-Escontrela, I; Arce, A.; Soto, A.; Marcilla, A.; Olaya, M.M.; Labarta, J.A. Correlation of Three-Liquid-Phase Equilibria involving Ionic Liquids. Physical Chemistry Chemical Physics (Royal Society of Chemistry). **2016**, 18, 21610-21617 (http://dx.doi.org/10.1039/C6CP03467E).

[25] Marcilla, A.; Labarta, J.A.; Olaya, M.M. Phase Equilibrium Correlation. Regarding the full consistency of the thermodynamics. 10th World Congress of Chemical Engineering. Barcelona (SPAIN), 1t-5th October, **2017**. RUA: <u>http://hdl.handle.net/10045/70035</u>.

[26] Marcilla, A.; Olaya, M.M.; Labarta, J.A.; P. Carbonell. Procedure for the correlation of normal appearance VLE data, where the classical models dramatically fail with no apparent

 Boundaries\_LL\_NRTL: GUI for the Characterization of the NRTL Model:

 Binary Spinodal Surfaces (in the τ<sub>i,j</sub>- τ<sub>j,i</sub>-x<sub>i</sub> space), LLE Maps, and Miscibility Boundaries

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 <u>http://hdl.handle.net/10045/121471</u>

reason. Fluid Phase Equilibria. **2019**, 493, 88-101 (https://dx.doi.org/10.1016/j.fluid.2019.04.001).

[27] M. Vicente-Martínez, J.A. Labarta. Análisis topológico del modelo NRTL en sistemas binarios y ternarios para caracterizar la tipología de los sistemas líquido-líquido que predice en función de los valores de los parámetros de interacción binaria. Institutional Repository of the University of Alicante (<u>RUA</u>) 2021. Available at: <u>http://hdl.handle.net/10045/117665.</u>

[28] Labarta, J.A.; Serrano, M.D.; Velasco, R.; Olaya, M.M.; Marcilla, A. Approximate Calculation of Distillation Boundaries for Ternary Azeotropic Systems. *Ind. Eng. Chem. Res.* **2011**, 50 (12), 7462-7466. DOI: <u>http://dx.doi.org/10.1021/ie101873</u>.

[29] Labarta, J.A.; Caballero, J.A.; Marcilla, A. Numerical Determination of Distillation Boundaries for Multicomponent Homogeneous and Heterogeneous Azeotropic Systems. *Computer Aided Chemical Engineering.* **2010**, 28(C), 643-648. DOI: <u>http://dx.doi.org/10.1016/S1570-7946(10)28108-7</u>. *Institutional Repository* (<u>RUA</u>): <u>http://hdl.handle.net/10045/14203</u>.

[30] Labarta, J.A. Procedure to Qualitatively Represent Residue Curve Maps and Analyze the Possible Existence of Distillation Boundaries. *Institutional Repository* (RUA, 2023): http://hdl.handle.net/10045/131058.

[31] Labarta, J.A.; Caballero, J.A. **ParamIni\_LL\_NRTL**: Graphical User Interface (GUI) for the Selection of NRTL Initial Parameters for the Correlation of Ternary Liquid-Liquid Equilibrium Data (Type I, II, III and 0 (LL island), i.e. with 1, 2, 3 or 0 binary pairs partially miscible). Institutional Repository of the University of Alicante (<u>RUA</u>). **2022**. Available at: <u>http://hdl.handle.net/10045/130017</u>.



[32] Labarta, J.A.; Caballero, J.A.; Marcilla, A. Liquid-Liquid Equilibrium Data Correlation: Predicting a robust and consistent set of initial NRTL parameters. Computer Aided Chemical Engineering (ESCAPE 33). **2023**, 52, 3457-3462. DOI: <u>https://doi.org/10.1016/B978-0-443-15274-0.50552-7</u>. Inst. Rep. of the Univ. of Alicante (<u>RUA</u>) 2023: http://hdl.handle.net/10045/134753.

[33] Labarta, J.A. Review of Graphical User Interfaces (GUIs) related to the Correlation of Phase Equilibrium Data. 2023. DOI: <u>https://doi.org/10.5281/zenodo.7628814</u>.

[34] Sapkowski, M.; Hofman, T. Problems and Limitations in the Calculation Of Liquid-Liquid Equilibrium. Fluid Phase Equilibria, **2023**, 571, 113823.

[35] Marcilla, A.; Gómez, A., García, A.N.; Beltrán, M.I.; Olaya, M.M.; Labarta, J.A. Operaciones de separación de transferencia de materia. <u>Editorial Síntesis (2022)</u>. ISBN: 9788413571775.

[36] Labarta, J.A. Procedure to Qualitatively Represent Residue Curve Maps and Analyze the Possible Existence of Distillation Boundaries (RUA, **2023**). <u>http://hdl.handle.net/10045/131058</u>.



[37] Gómez, A.; Ruiz, F.; Marcilla, A.; Labarta, J.A.; Menargues, S. Diseño de la separación de mezclas ternarias (I). Conceptos gráficos del equilibrio entre fases. Ingeniería Química. 2001, 377, 219-229. Institutional Repository (RUA): http://hdl.handle.net/10045/24715.

[38] Gómez, A.; Ruiz, F.; Marcilla, A.; Labarta, J.A.; Menargues, S. Diseño de la separación de mezclas ternarias (II). Aplicación de conceptos gráficos a la separación de mezclas azeotrópicas. Ingeniería Química. 2001, 379, 253-262. Institutional Repository (<u>RUA</u>): <u>http://hdl.handle.net/10045/24716</u>.

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